

Leaving Group Ability

Leaving Group	pK _b	Comment
I ⁻	-10	Excellent to good leaving groups
N ₂	<-10	
TfO ⁻	<-10	
Br ⁻	-9	
ArSO ₃ ⁻	-7	
Cl ⁻	-7	
CF ₃ CO ₂ ⁻	0.2	
H ₂ PO ₄ ⁻	2.2	

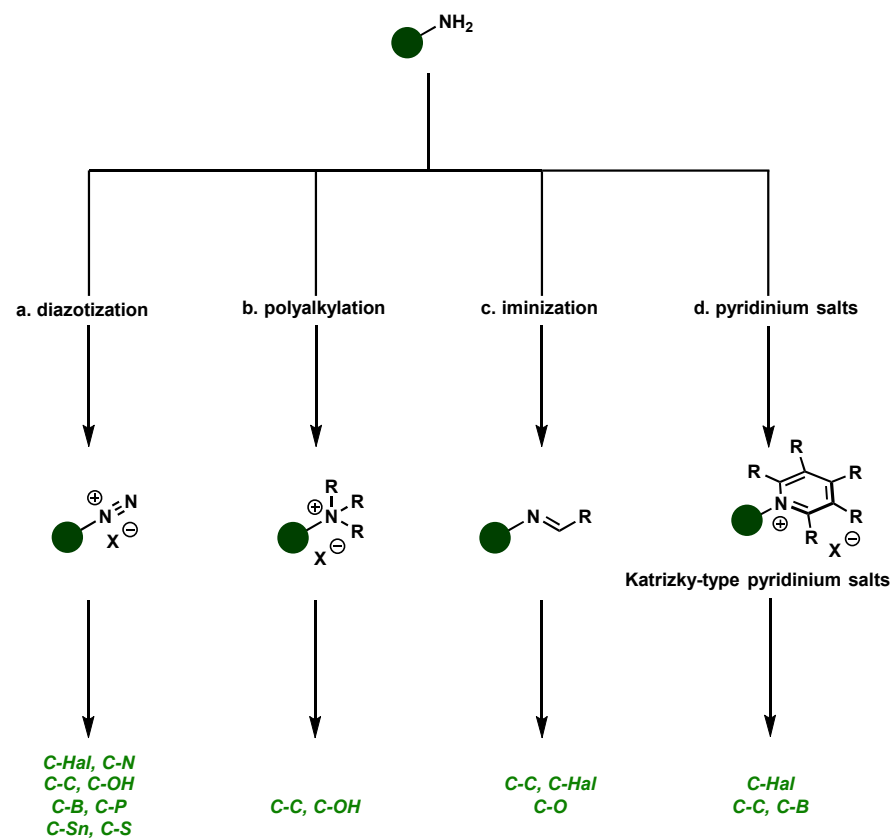
F ⁻	3.2	Fair to poor leaving groups
CH ₃ CO ₂ ⁻	4.8	
CN ⁻	9.1	
NH ₃	9.2	
RNH₂, R₂NH, R₃N	10	
CH ₃ CH ₂ S ⁻	10.6	
CH ₃ O ⁻	15.5	
HO ⁻	15.7	
CH ₃ CH ₂ O ⁻	15.9	
(CH ₃) ₃ CO ⁻	18	

H₂N⁻	36	Not leaving groups
CH ₃ ⁻	49	

Leaving group ability is related with its pK_b.

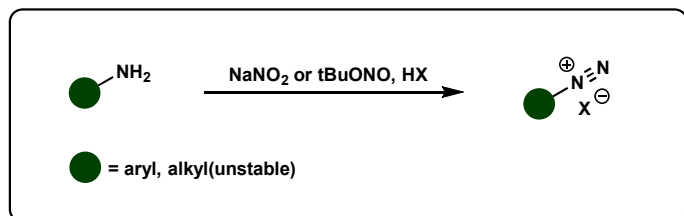
The less basic the leaving group is, the easier it is to dissociate.

Amines activation strategies

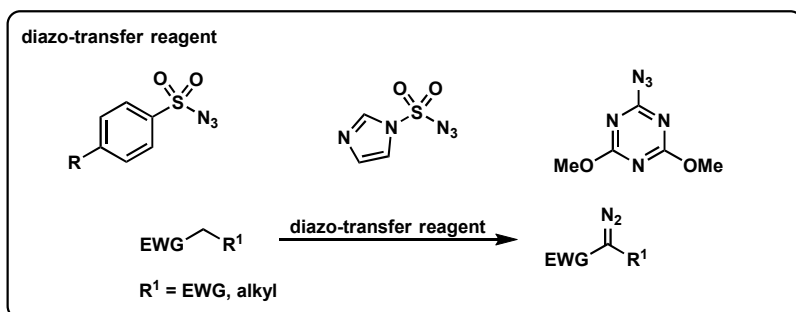


Synthesis of diazo compounds

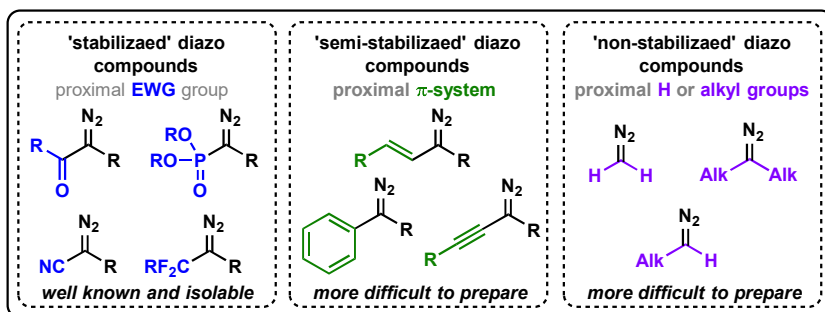
Diazotization of amine



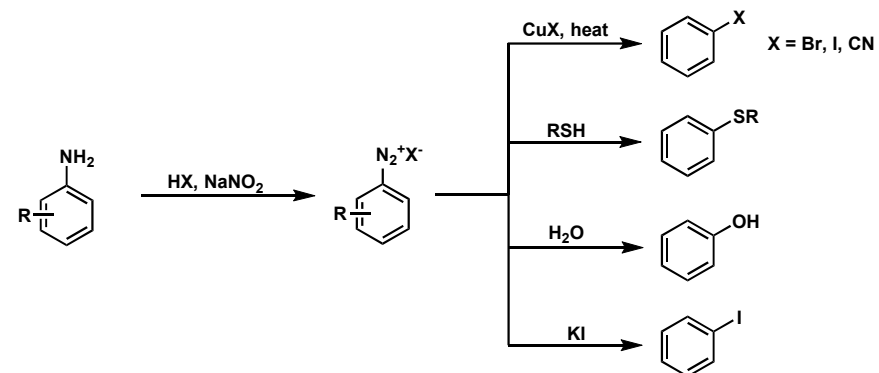
Prepare diazo compounds from diazo-transfer reagent



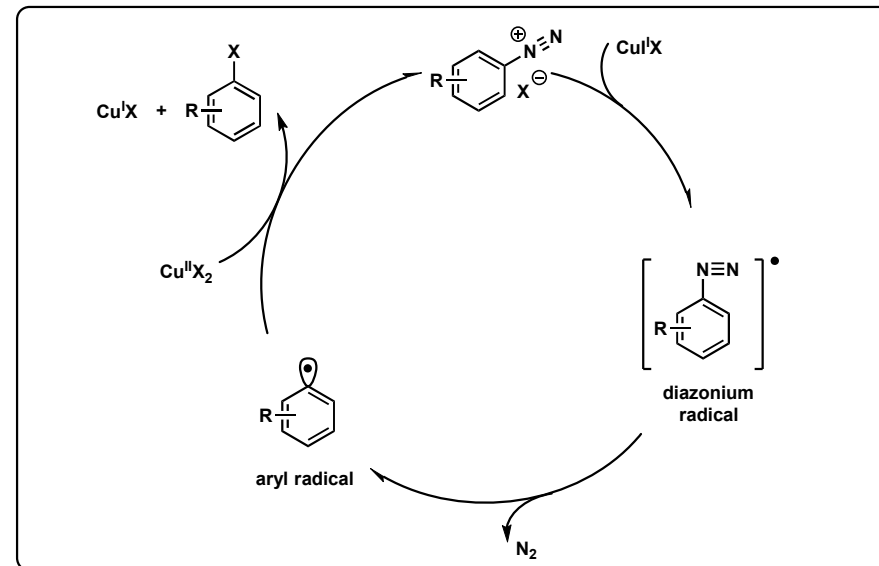
Stability of diazo compounds



Sandmeyer reaction (1884)

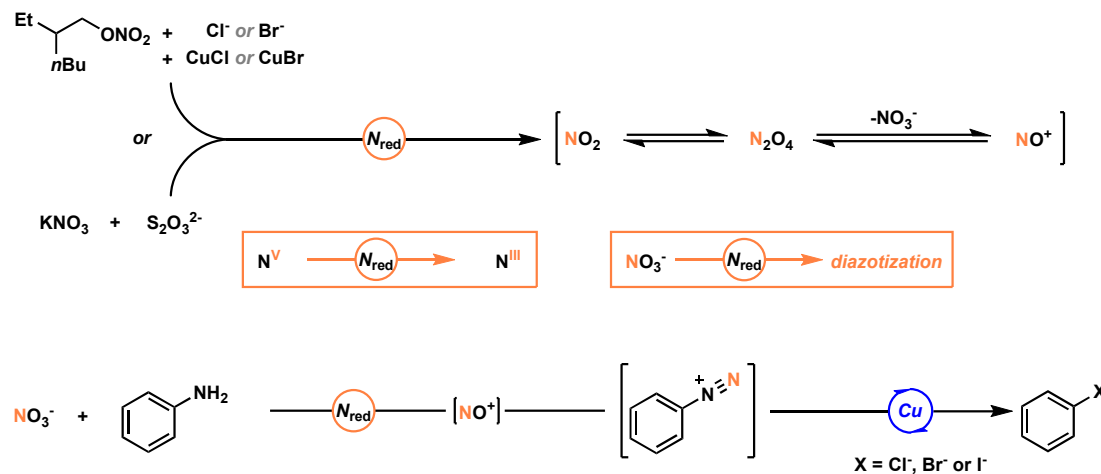


Mechanism:

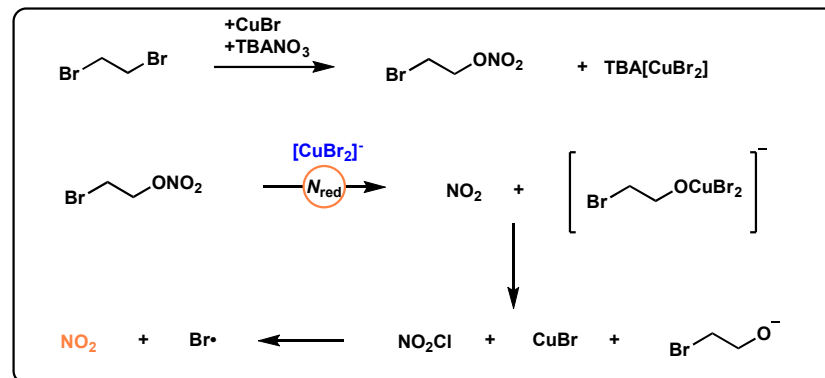
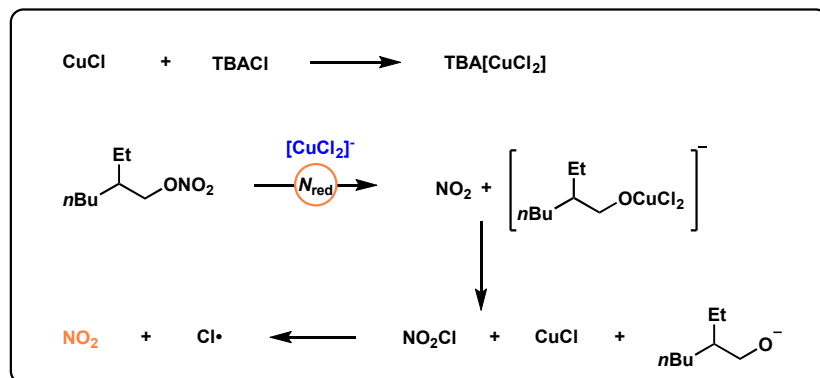


The mechanism of the Sandmeyer reaction is not completely understood.

Safer Sandmeyer Reaction



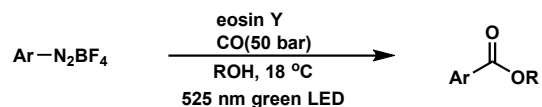
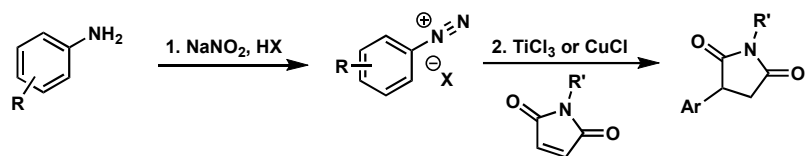
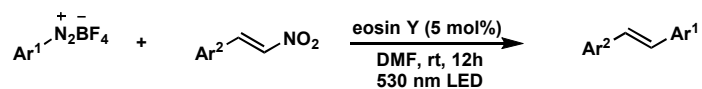
Mechanism:



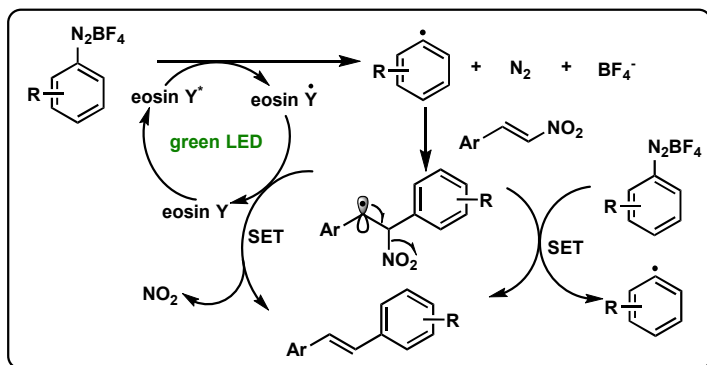
Ritter et al., Science 2024, 384, 446

C-C Bond Formations

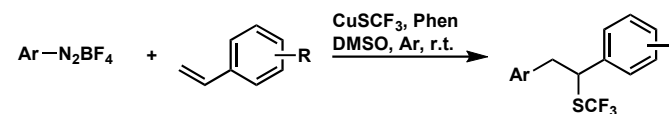
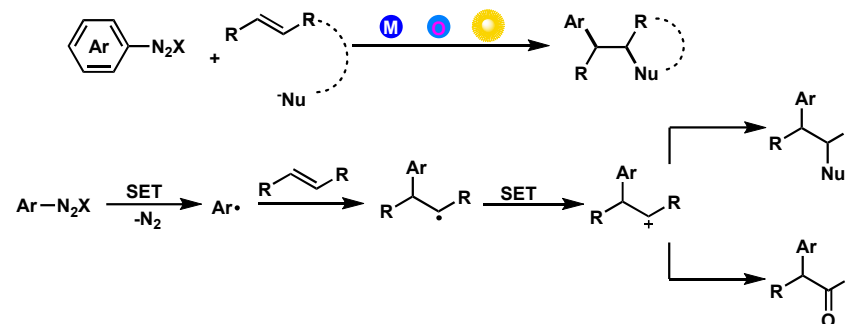
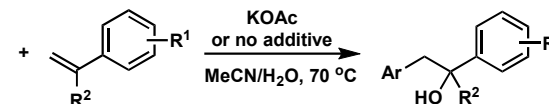
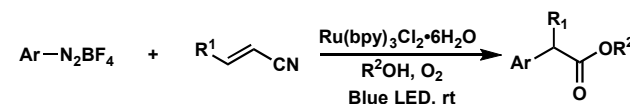
1. ArylRadical Intermediates

Wangelin et al., *Angew. Chem., Int. Ed.* 2015, 54, 2270Zhao et al., *RSC Adv.* 2016, 6, 23438Wang et al., *Chem. Commun.* 2016, 52, 14234

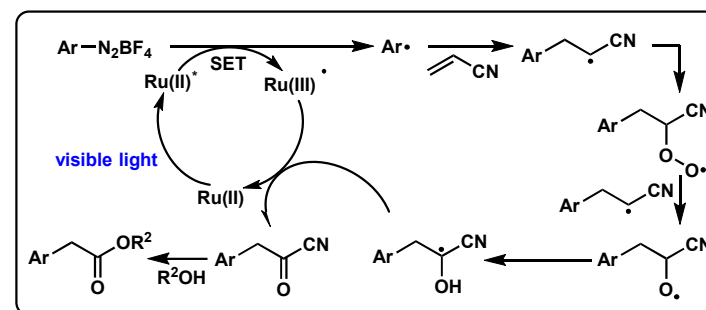
mechanism:



Difunctionalization

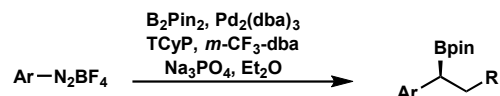
Chen et al., *J. Org. Chem.* 2018, 83, 5836Heinrich et al., *Angew. Chem., Int. Ed.* 2016, 55, 8744

mechanism:

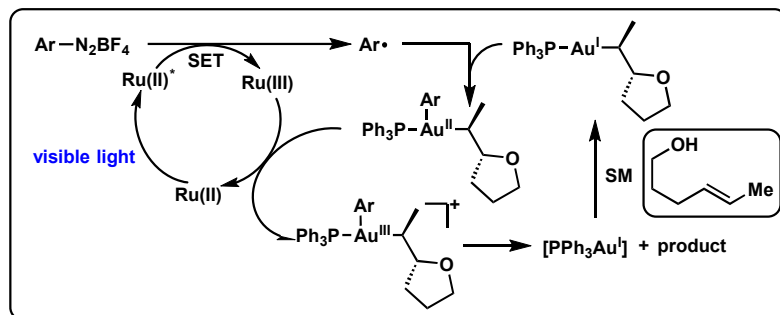
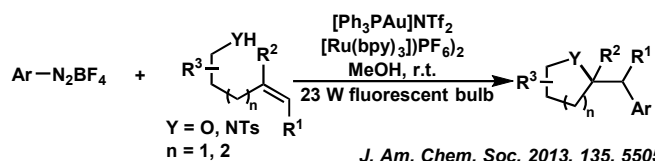
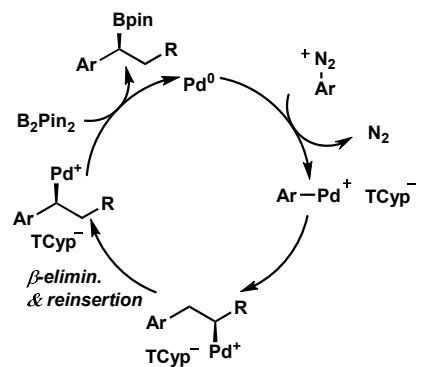
Jiang et al., *Eur. J. Org. Chem.* 2015, 2015, 5775

C-C Bond Formations

2. Transition-Metal-Catalyzed

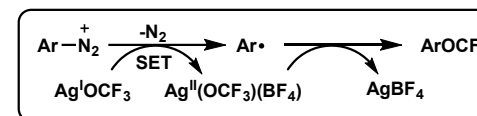
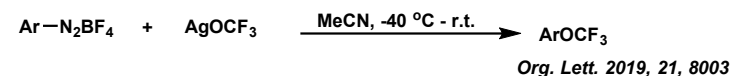


J. Am. Chem. Soc. 2015, 137, 3213

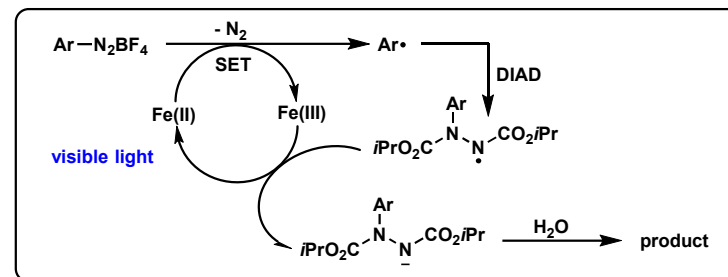
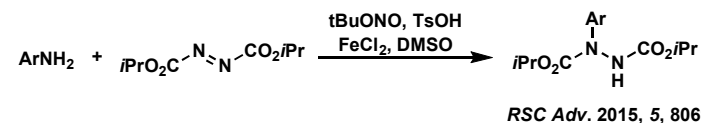


C-X Bond Formations

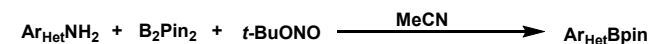
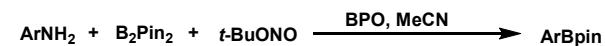
1). C-O Formations



2). C-N Formations



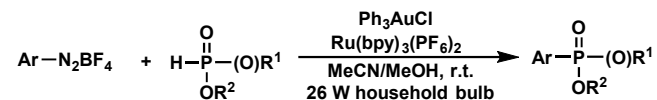
3). C-B Formations



Angew. Chem., Int. Ed. 2010, 49, 1846

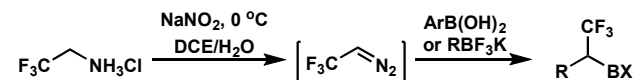
C-X Bond Formations

4). C-P Bond Formations



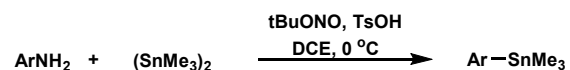
Chem. Sci. 2015, 6, 1194

Alkyl diazo compounds

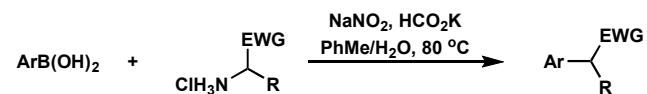


Angew. Chem. Int. Ed. 2013, 52, 13656.

5). C-Sn Bond Formations



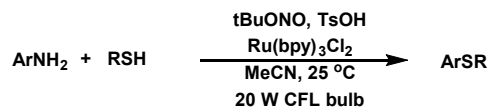
Angew. Chem., Int. Ed. 2013, 52, 11581



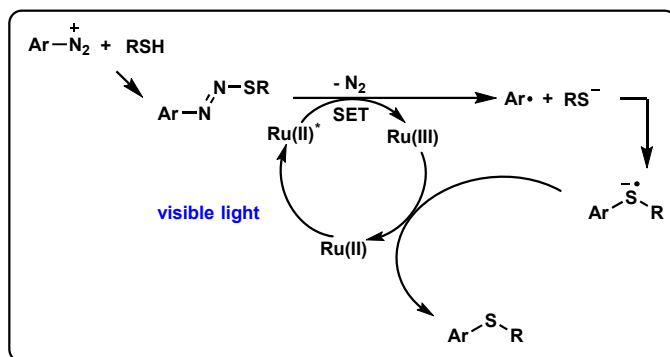
R = alkyl, EWG = CO₂R', CN

Angew. Chem. Int. Ed. 2014, 53, 10510.

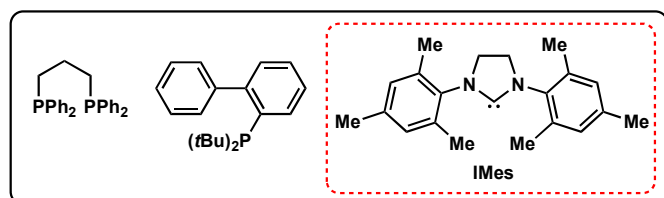
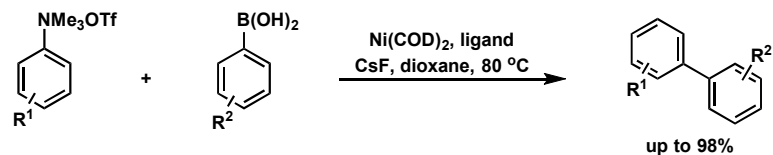
6). C-S Bond Formations



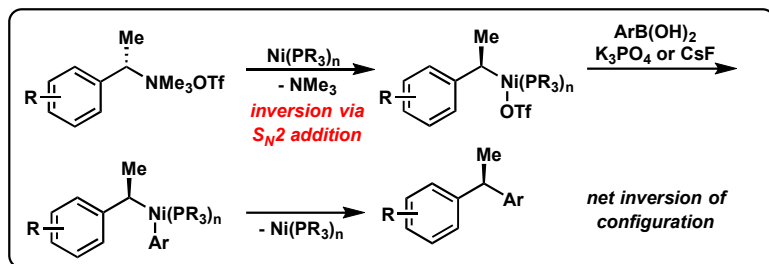
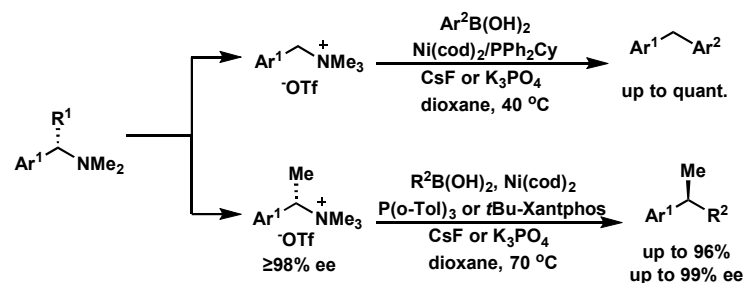
Angew. Chem., Int. Ed. 2013, 52, 7860



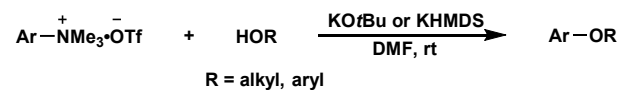
Quaternary Ammonium Salt



J. Am. Chem. Soc. 2003, 125, 6046

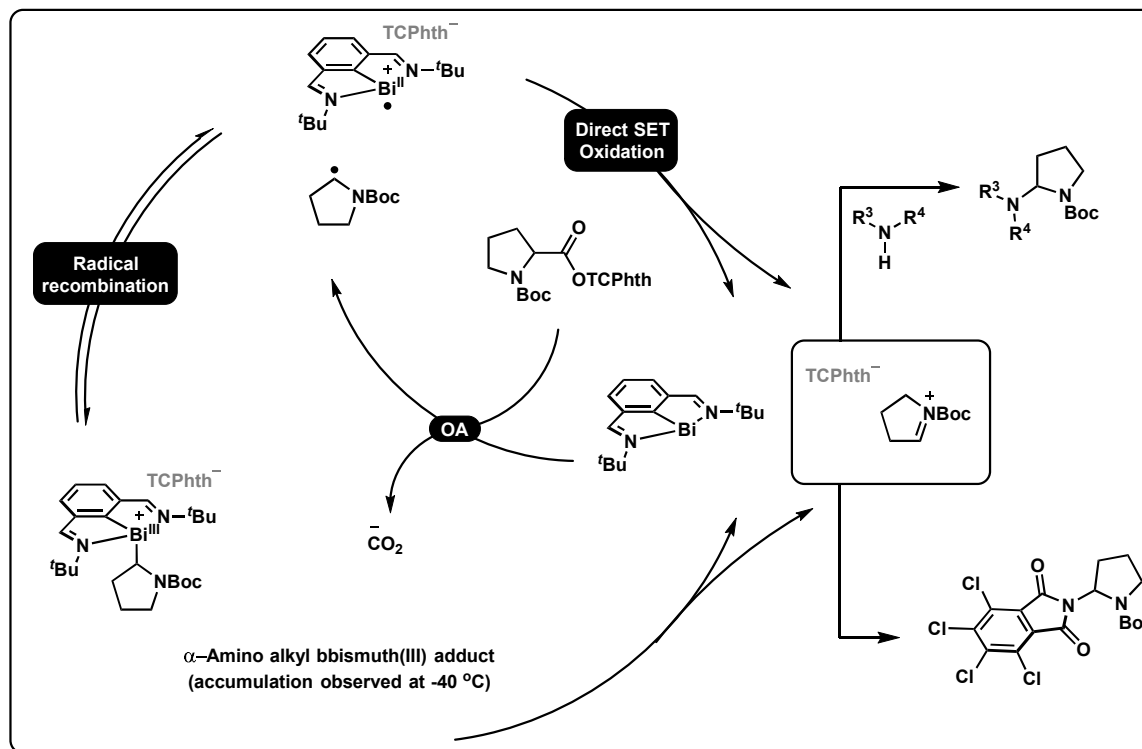
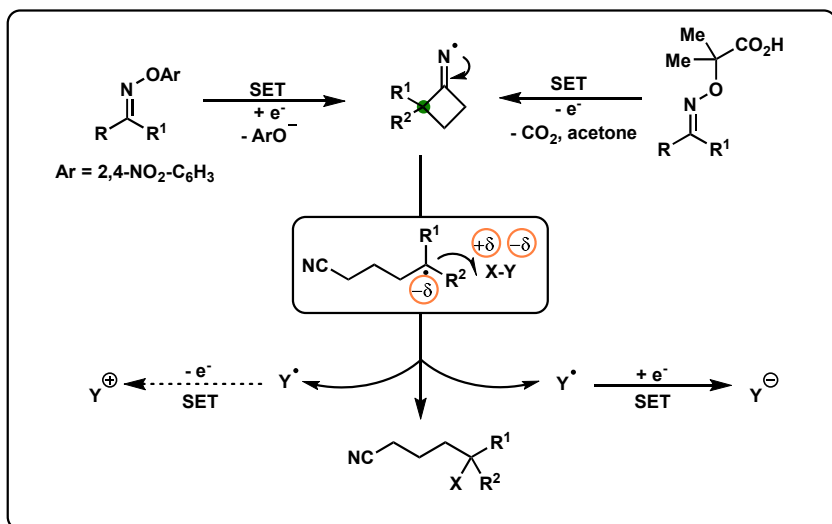
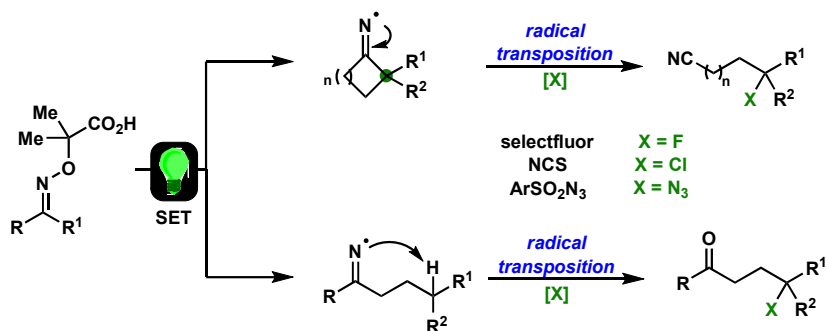
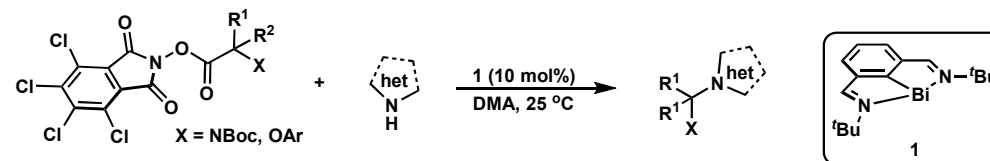
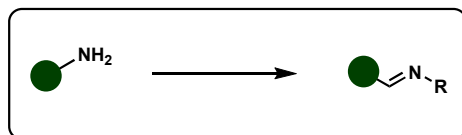


J. Am. Chem. Soc. 2013, 135, 280



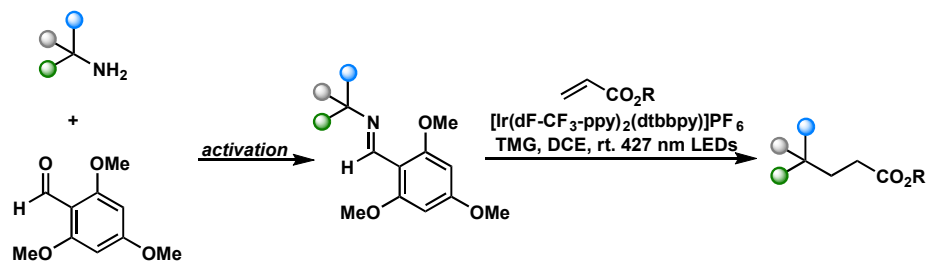
Angew. Chem. Int. Ed. 2018, 57, 3641

Imine Derivatives

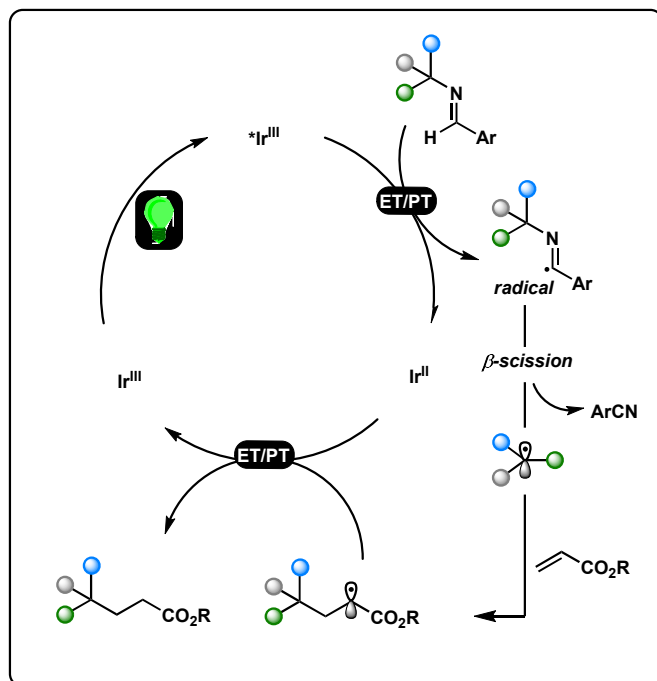
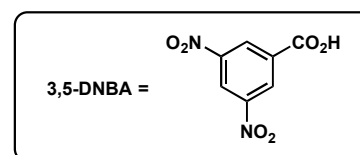
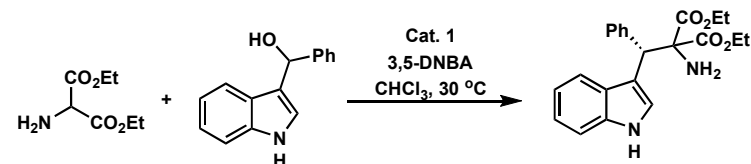


Nat. Chem. 2023, 15, 1138

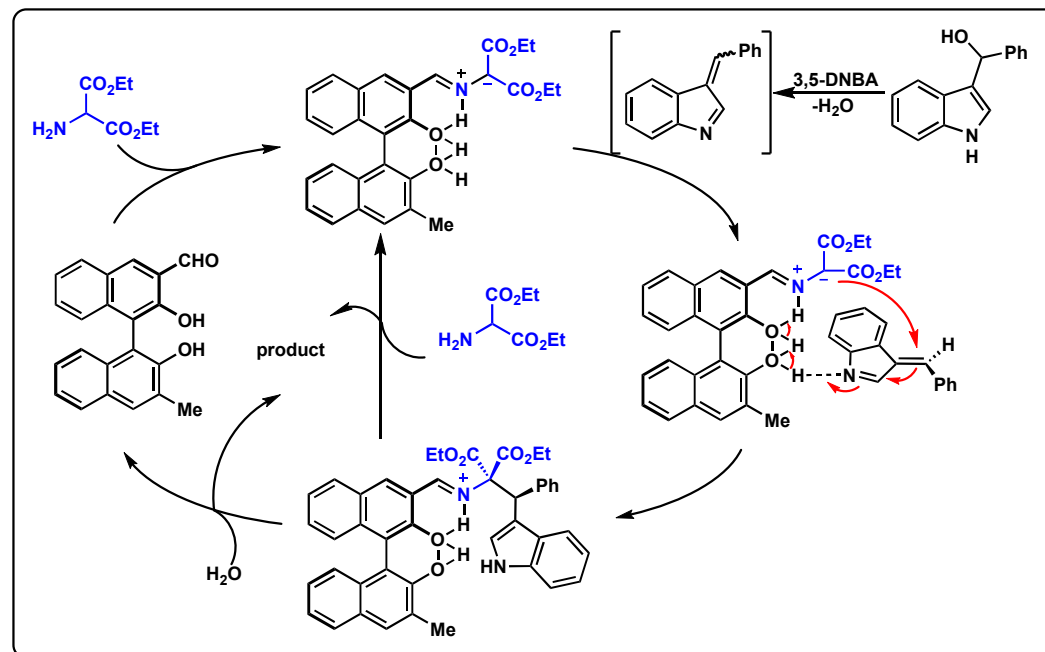
Imine Derivatives



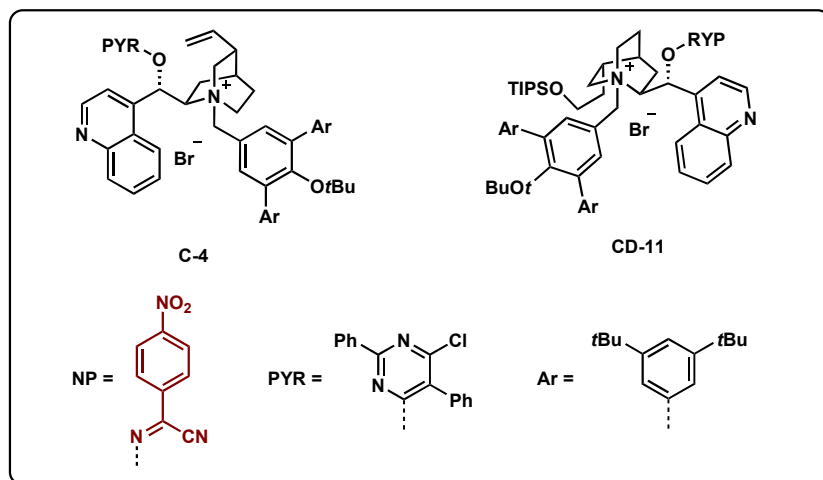
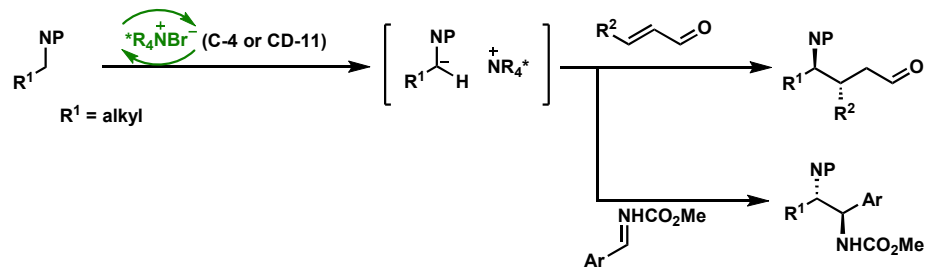
J. Am. Chem. Soc. 2020, 142, 18310



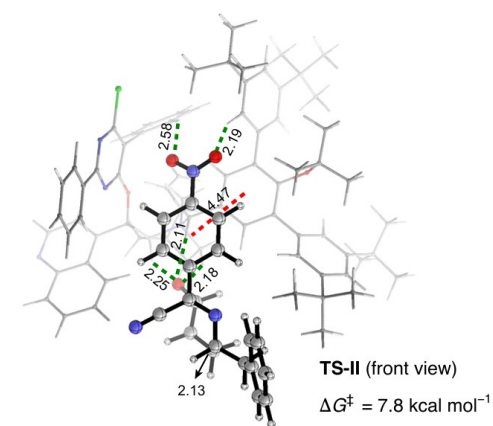
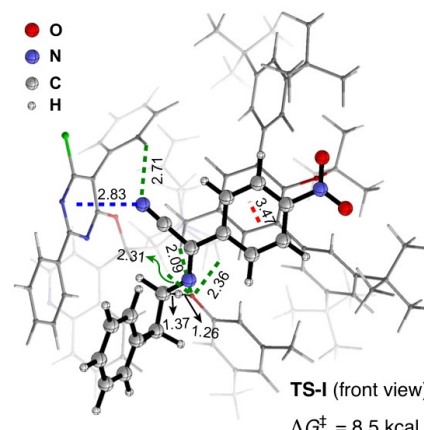
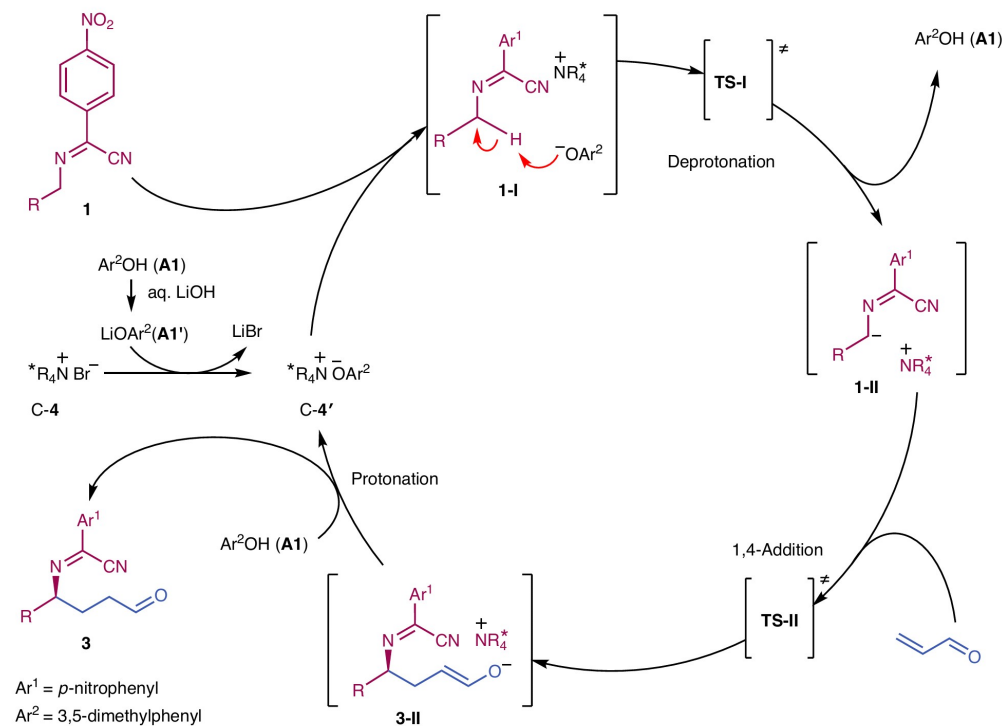
Chem. Sci. 2014, 5, 1988



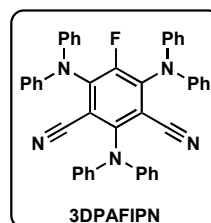
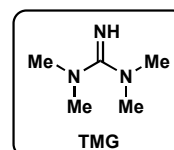
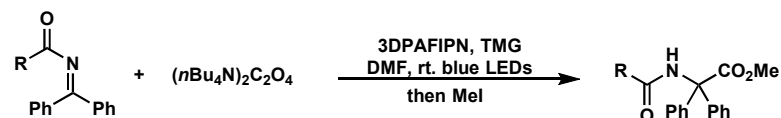
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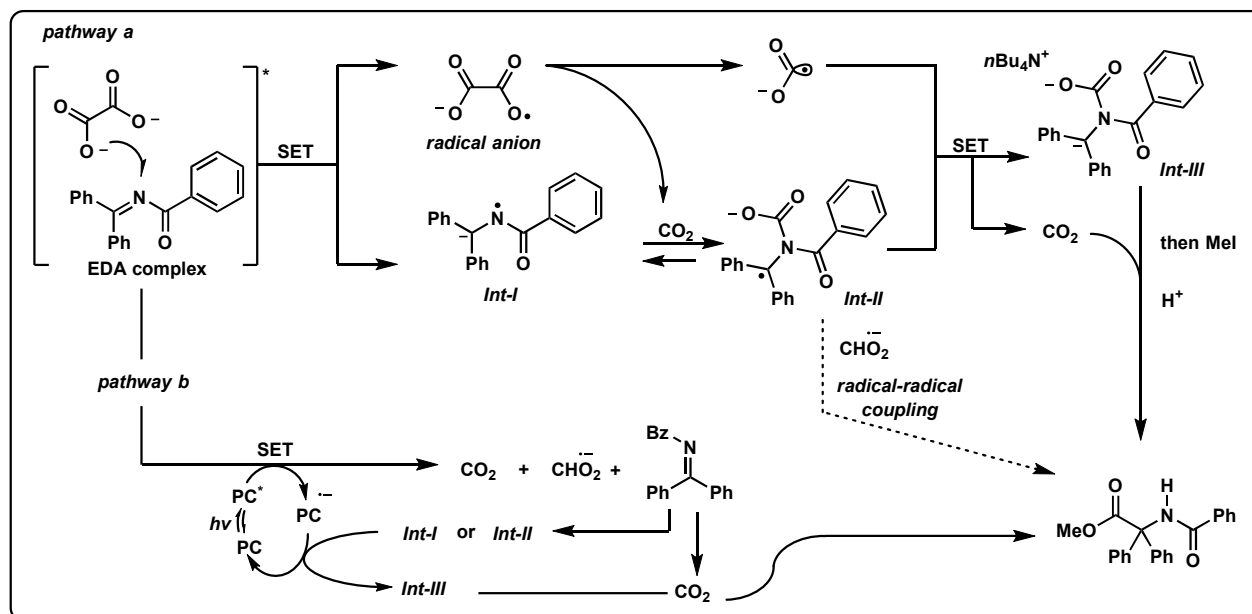
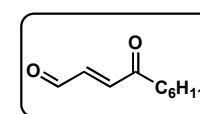
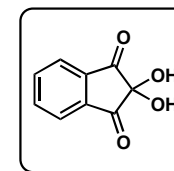
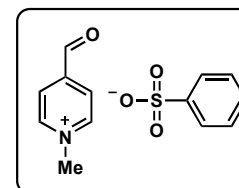
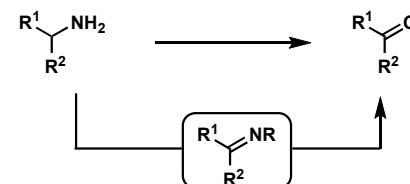
Nat. Catal., 2024, 7, 1076



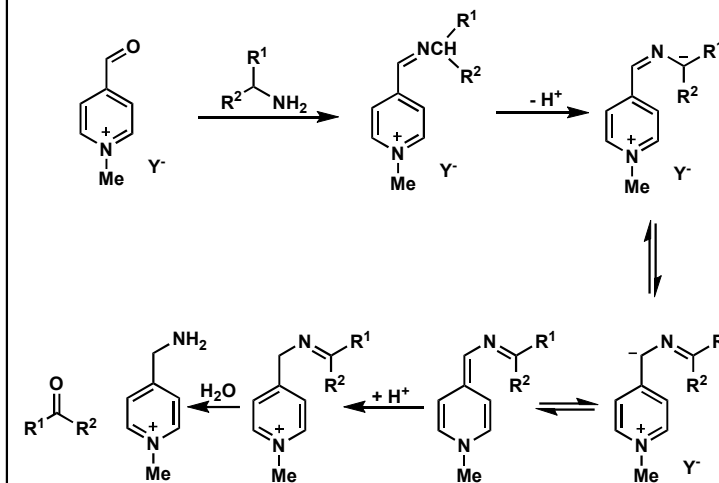
Imine Derivatives



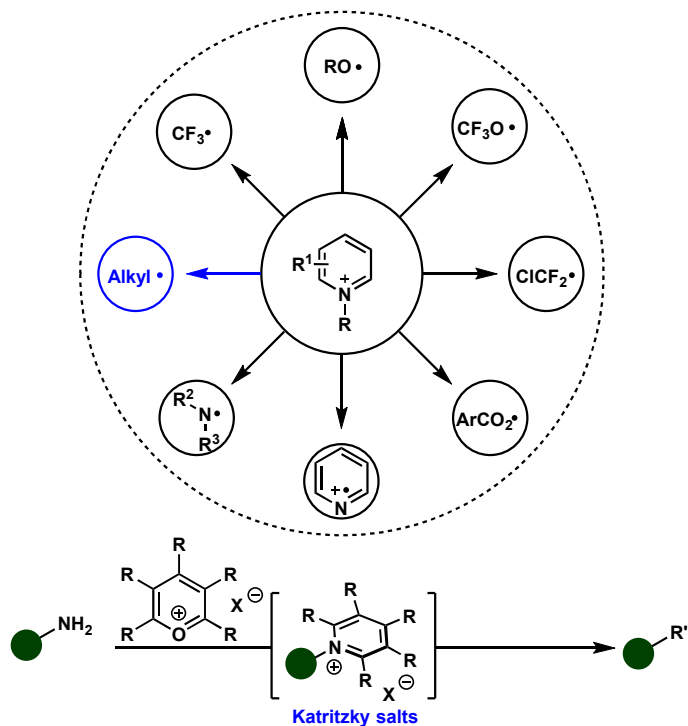
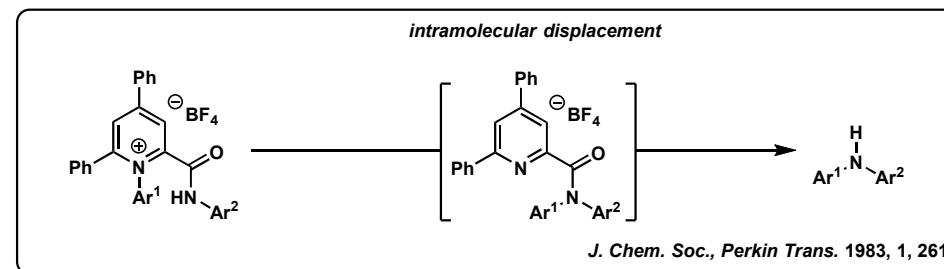
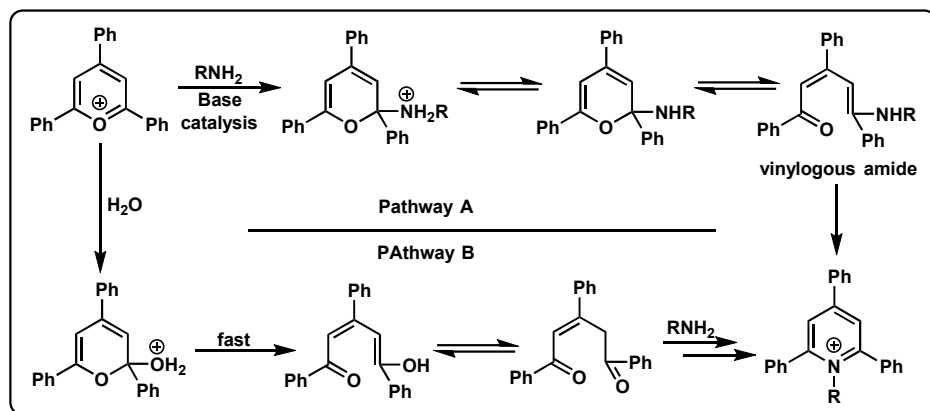
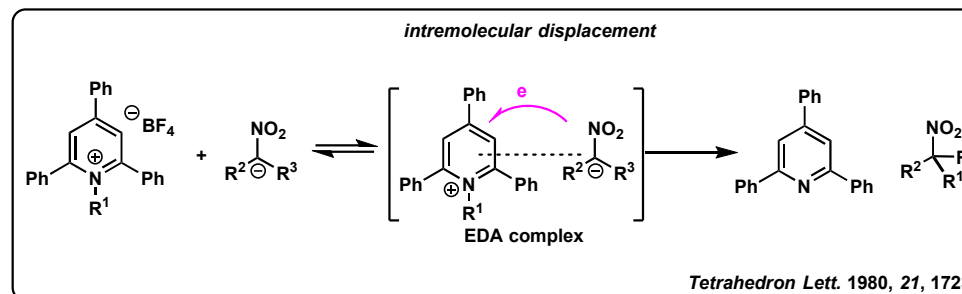
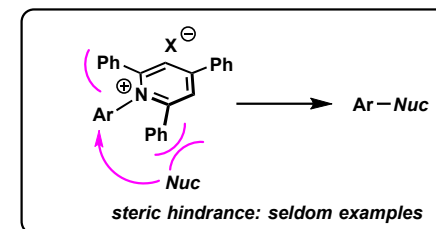
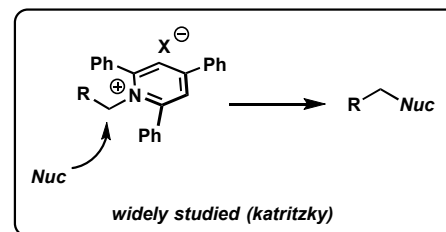
Transfer to Ketone(Aldehyde)



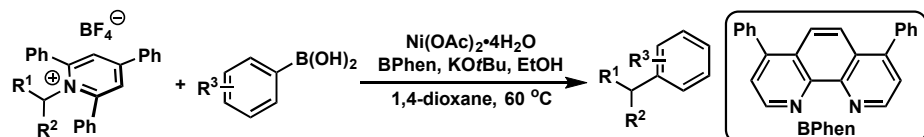
ACS Catal. 2024, 14, 10053



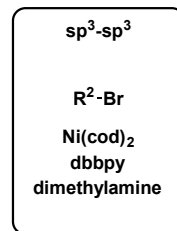
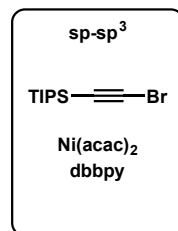
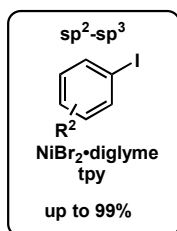
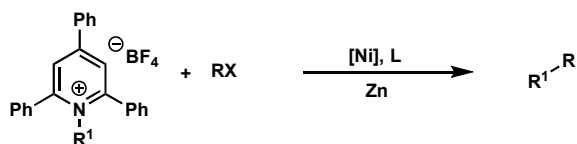
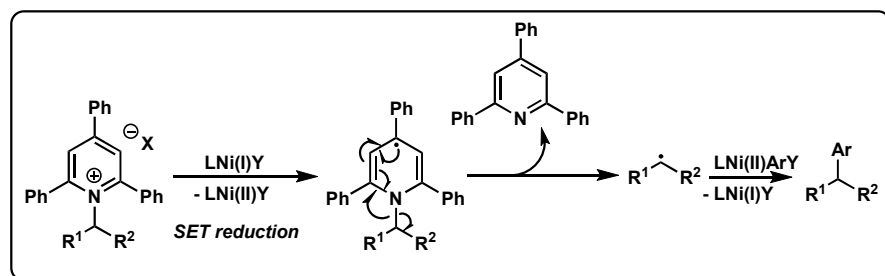
Pyridinium Salts

SN₂ or SN₁ Type

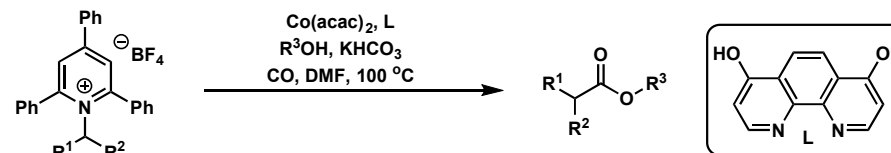
Metal-Catalyzed Cross-Couplings



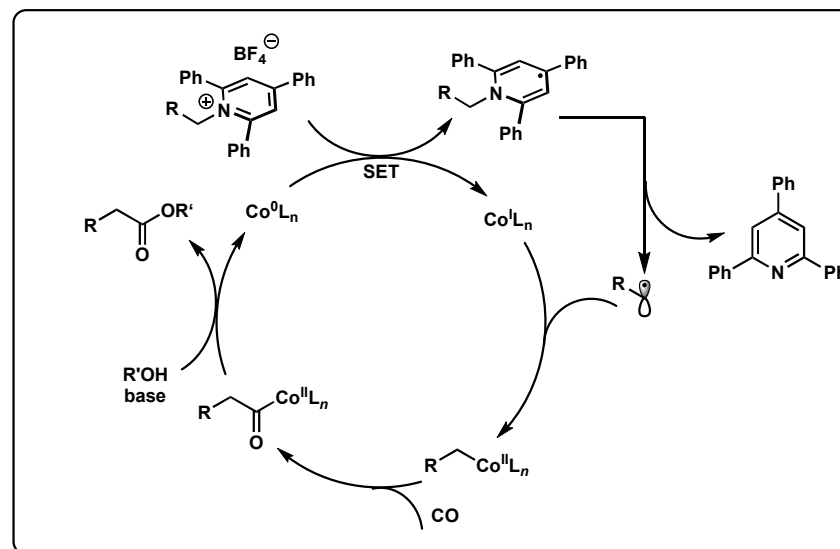
J. Am. Chem. Soc. 2017, 139, 5313



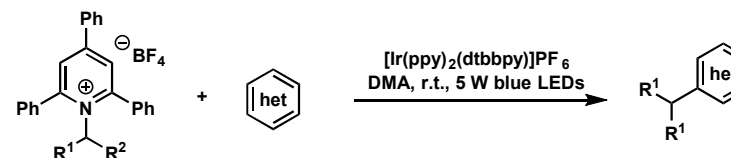
Sci. Adv. 2019, 5, 9516.



Org. Lett. 2019, 21, 6919

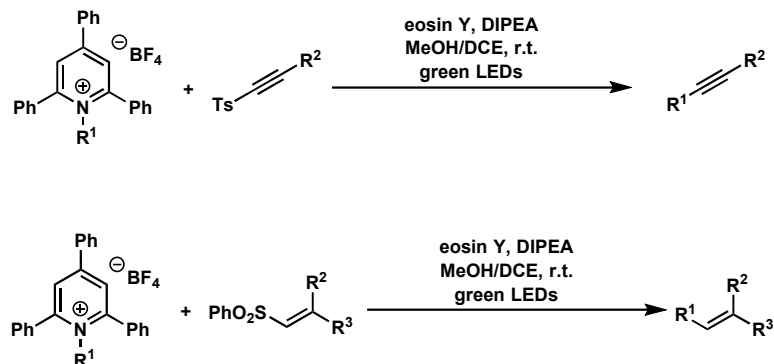


Photoinduced Reactions C–C Bond Constructions



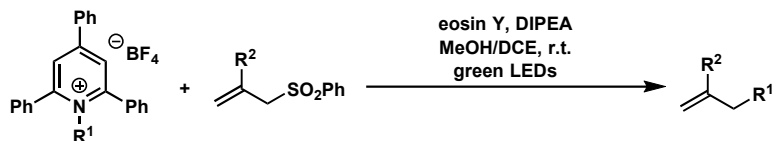
Angew. Chem. Int. Ed. 2017, 56, 12336

Desulfonylative Alkynylation/Alkenylation



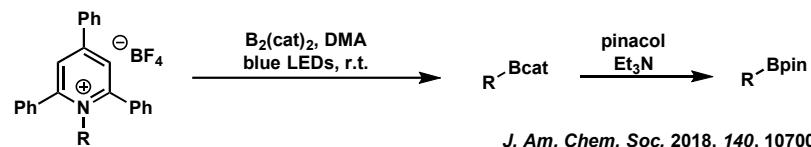
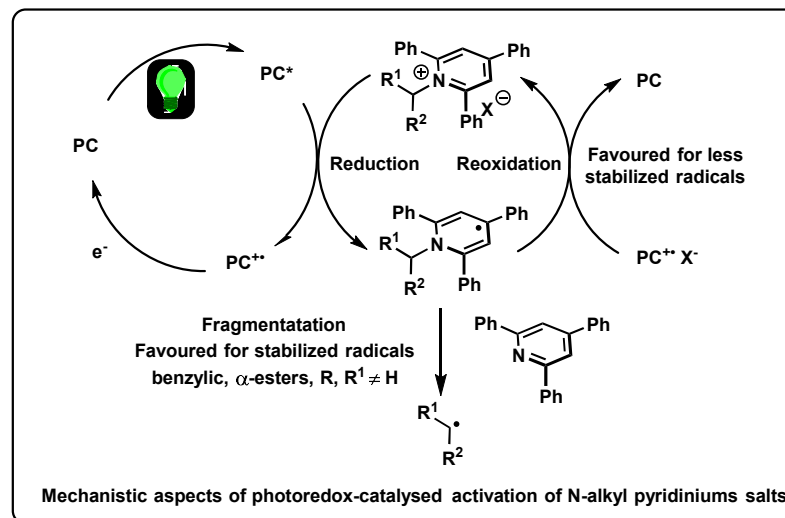
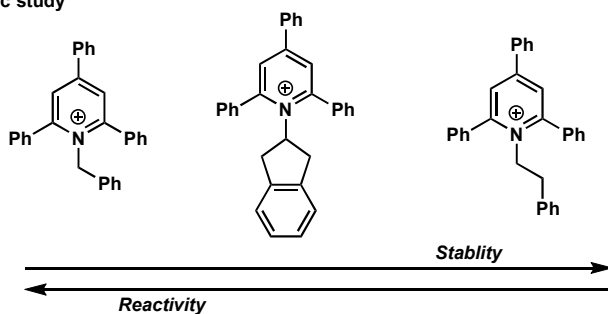
ACS Catal. 2018, 8, 11362

Desulfonylative Allylation



Angew. Chem. Int. Ed. 2019, 58, 5697

Kinetic study



J. Am. Chem. Soc. 2018, 140, 10700

