



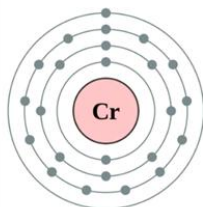
IUPAC: Chromium Atom number: 24

Oxidation state: II, III, V, VI

m.p. 1857 °C b.p. 2672 °C density:

7.19 g/cm³

Soluble in dilute sulfuric acid, dilute hydrochloric acid, insoluble in water, nitric acid, aqua regia

**History:**

1. In 1766, Lehmann, a professor of chemistry in St. Petersburg, Russia, analyzed chromium and determined that it contained lead.

2. In 1919, the structure of the resulting bis(benzene)chromium species was determined of the structure of the resulting bis(benzene)chromium species.¹



3. In 1957, Anet and Leblanc were the first to prepare aqueous solutions of benzylchromium compounds by reaction of Cr(II) with benzyl chloride.

4. Nozaki and Hiyama et al. screened of the reactivity of benzylchromium and related reagents in 1977 that triggered an explosive development of this particular branch of organometallic chemistry.

5. Takai, Nozaki and Hiyama found that organochromium species can add selectively to an aldehyde moiety without affecting the ketone or cyano group.

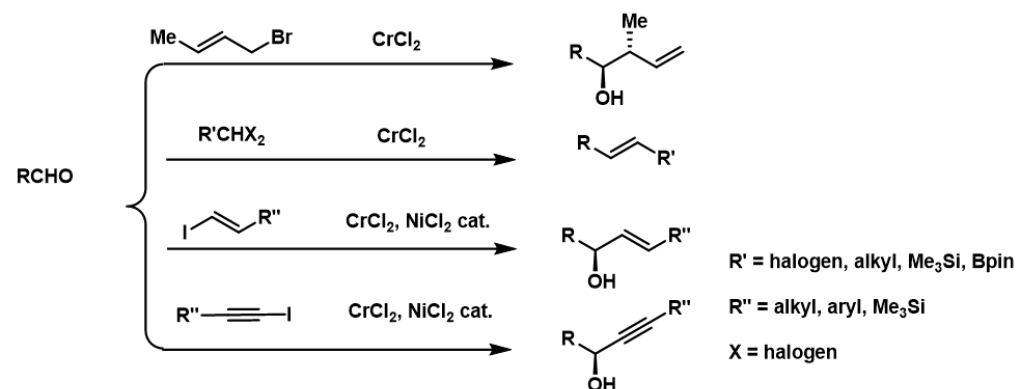
6. In 1986, Kishi et al. and Nozaki et al. discovered that traces of nickel salts exert a catalytic effect on the formation of the C-Cr(III) bond.

Furstner, Chem. Rev. 1999, 99, 991

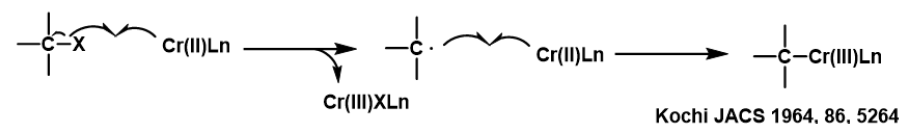
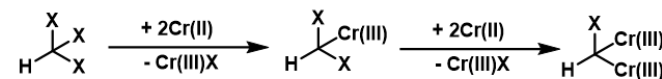
Preparation:

1. transmetalation from RLi, RZnX, RMgX
Cr(III) low solubility in ethereal solvents

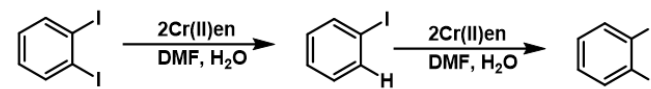
2. reduction of C-X

**Reduction of organic halides**

(X = I > Br > Cl)

**Polyhalides:**

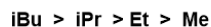
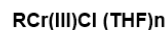
Takai JACS, 1986, 108, 7408

Aryl halides:

NR in aprotic solvents
Takai JACS 1986, 108, 6048

The nature of carbon-chromium bonds

1. thermal stability

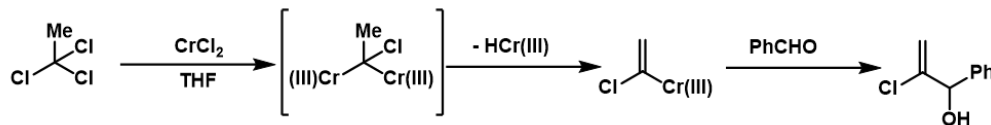


Ikeda J. organomet. Chem 1972, 37, 317.

Decylchromium dichloride

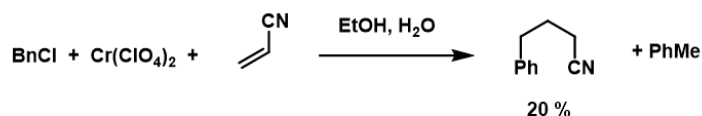
0 °C stable
20 °C slow homolysis
65 °C rapid homolysis

Zeiss J. organomet. Chem 1971,26,101

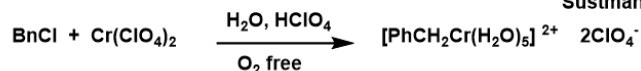


Miokoski JACS 2001, 123, 9196

2. hydrolysis

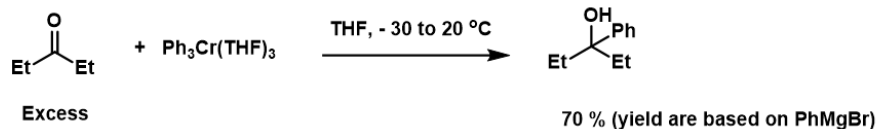


Sustmann chem. Ber 1993, 126, 1759.



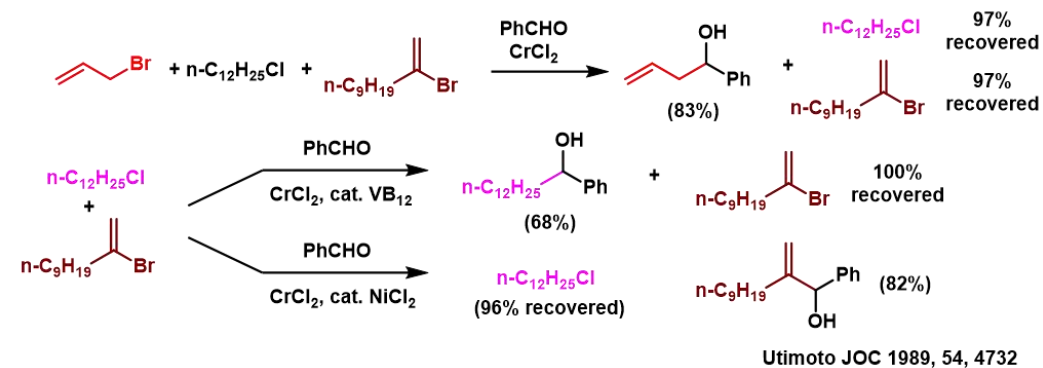
Anet Can. J. Chem. 1959, 37, 58

3. nucleophilicity

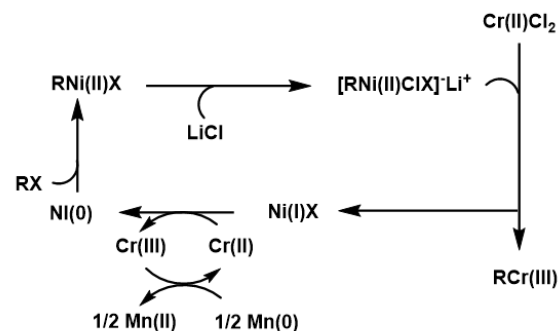


Zeiss J. organomet. Chem 1965, 4, 397

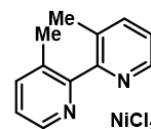
Chemoselectivity



Utimoto JOC 1989, 54, 4732

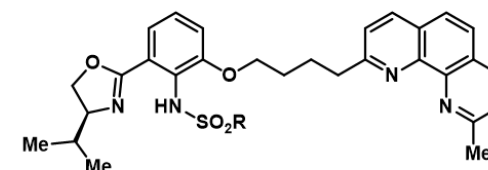
Alkenyl/Aryl chromium addition to aldehyde [NiCl₂, Pd(OAc)₂]"NiCl₂ and Pd(OAc)₂ have a dramatic effect"

Kishi JACS 1986, 108, 5644



Kishi OL 2004, 24, 5421

1 and 5 mol % catalyst-loading

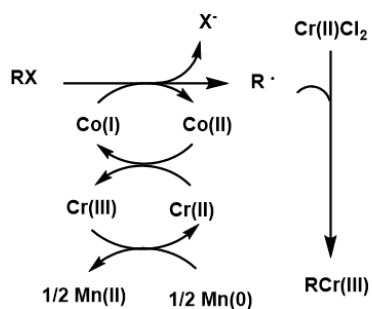
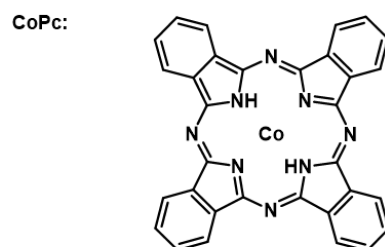
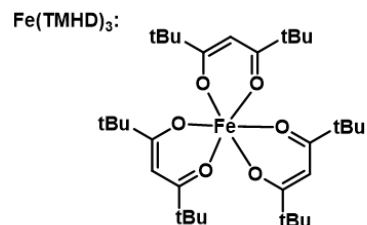
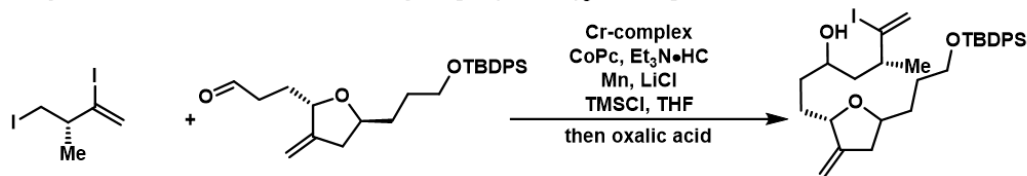


Kishi 2009,131,42, 15393

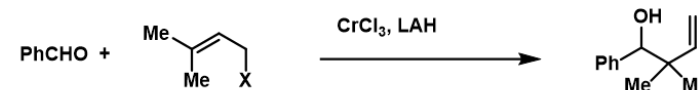
the molar ratio of coupling partners

PhI	PhCHO	85%	A
		62-88%	B
		66%	C

PhBr	PhCHO	31%	A
		57%	C

A:CrCl₂ (4 equiv), DMF, rt.B:CrCl₂ (15 mol %), Mn powder, chlorosilane, DMF/DME (20/3), 50 °C.C:CrCl₂ (10 mol %), Pd(OAc)₂ (0.1 mol %), PPh₃ (0.4 mol %), constant current conditions, LiClO₄ as supporting electrolyte in DMFAlkyl chromium addition to aldehyde [Fe(TMHD)₃, CoPc]Kishi OL 2004, 4, 25, 4435
Kishi JACS 2004, 126, 39, 12248

Allyl chromium addition to Aldehyde:

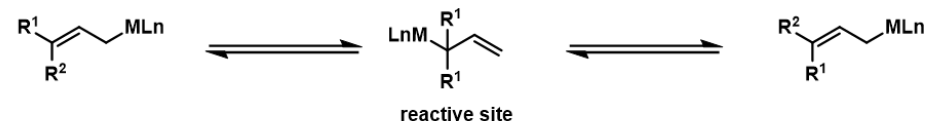
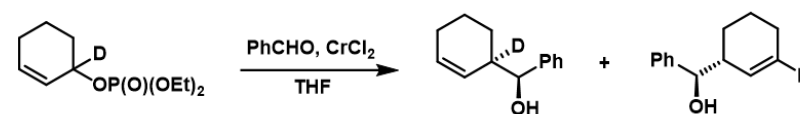


X = Br, THF, rt 90%

X = OP(O)(OEt)₂, DMF, 90 °C, 97%

Nozaki BCSJ 1982, 55, 561.

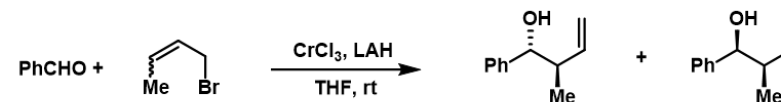
Regioselectivity:

Equilibrium rate: AllyLi, AllyMgBr > AllyBR'₂

1:1

Takai JSOJ, 1988, 46, 66.

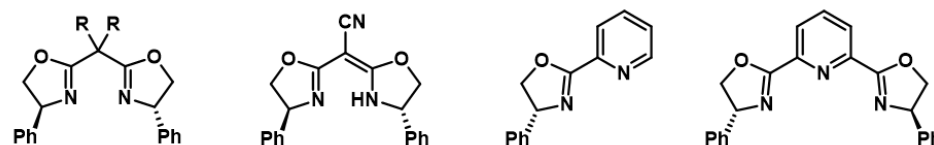
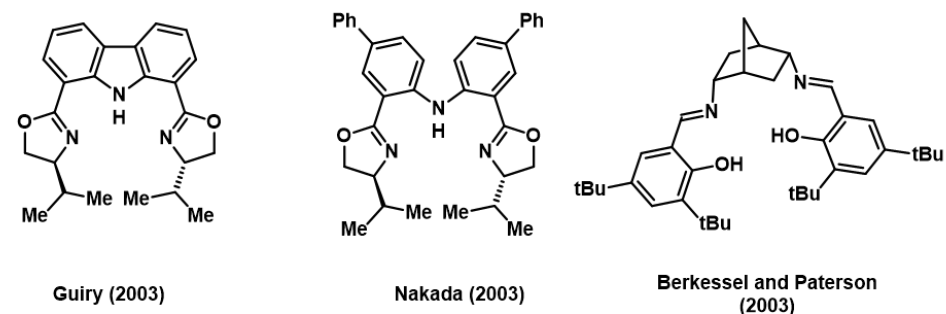
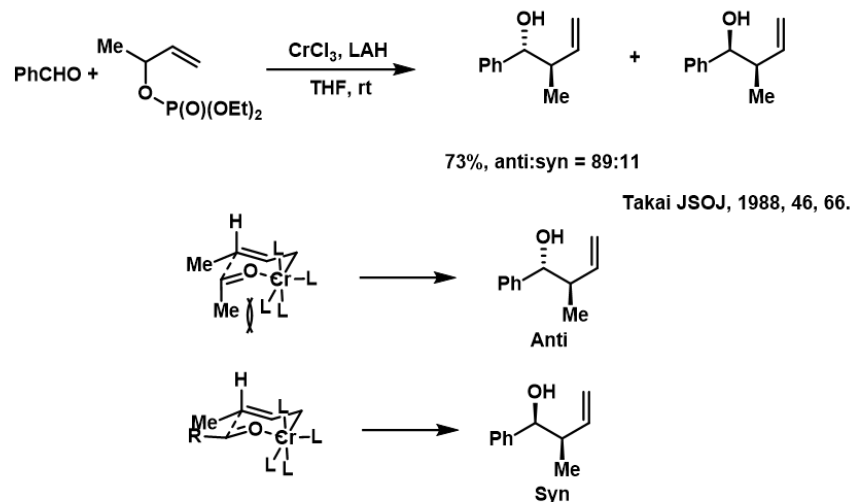
Diastereoselectivity:



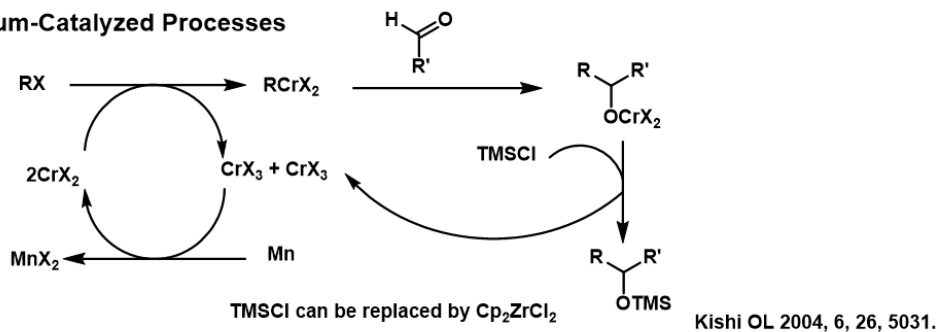
(E)-crotyl bromide 96%, anti:syn = 100:0

(Z)-crotyl bromide 87%, anti:syn = 100:0

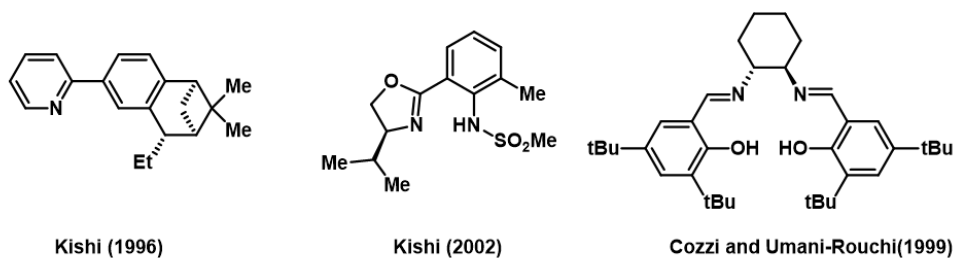
Heathcock, TL 1978, 1685



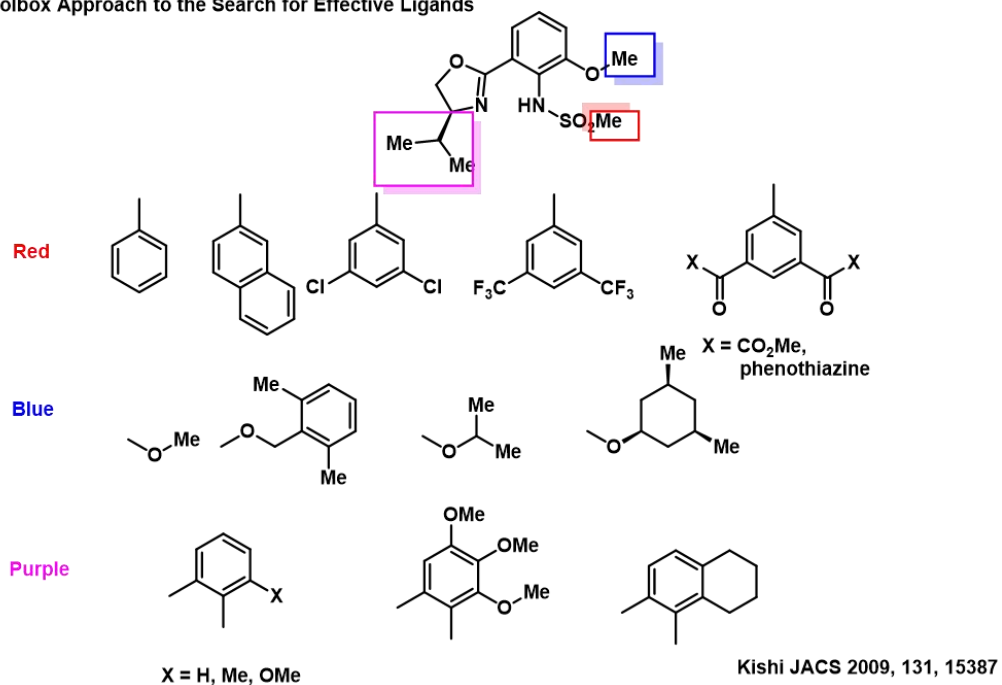
Chromium-Catalyzed Processes

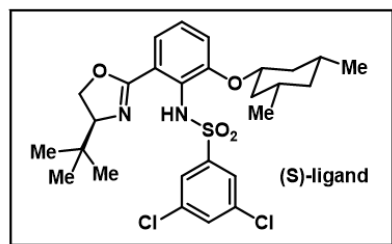
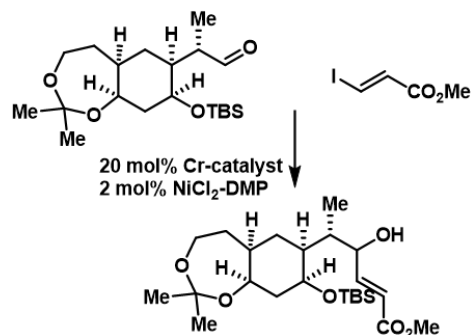


Asymmetric addition

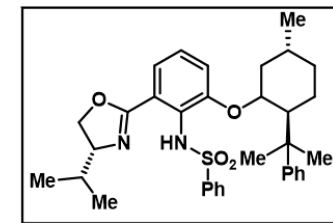
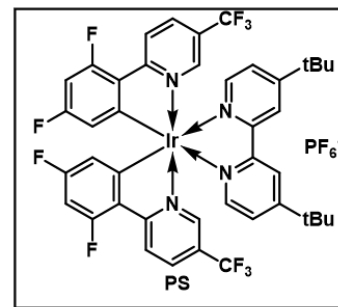
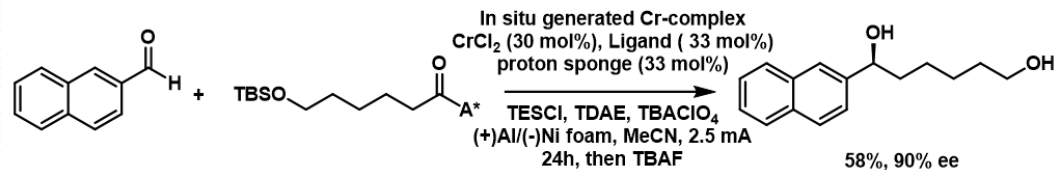


Toolbox Approach to the Search for Effective Ligands



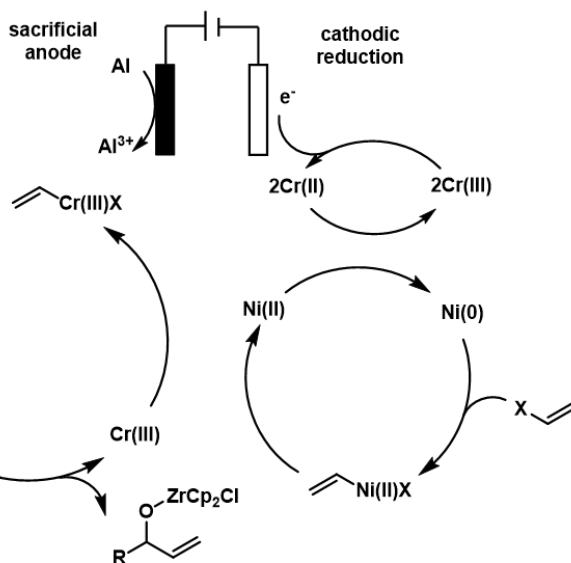
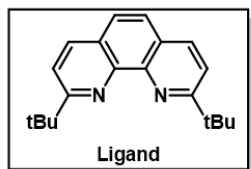
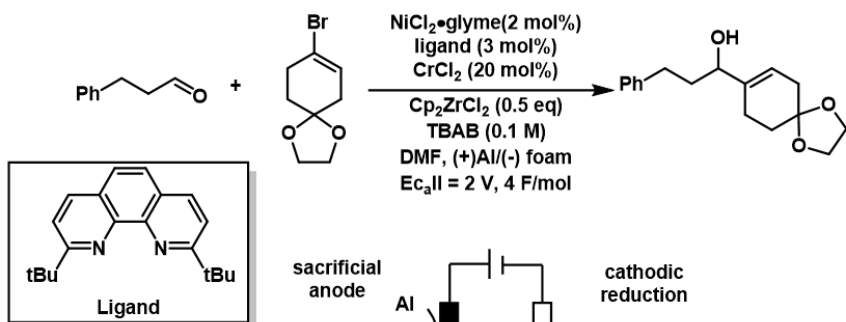


non-asymmetric ligands 1.1:1
(S)-ligand dr 60:1
(R)-ligand dr 1:27

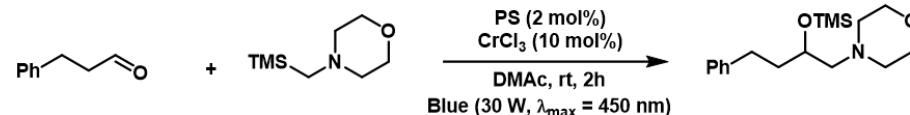


Baran JACS 2024, 146, 4872

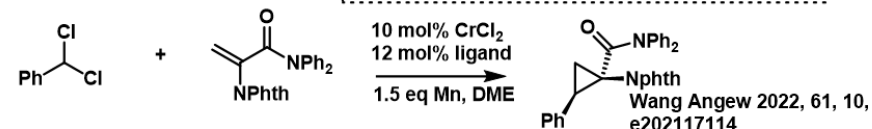
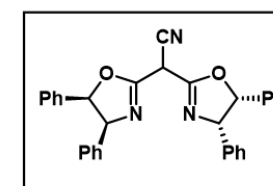
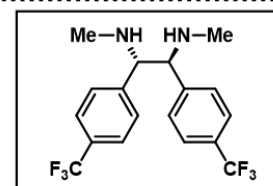
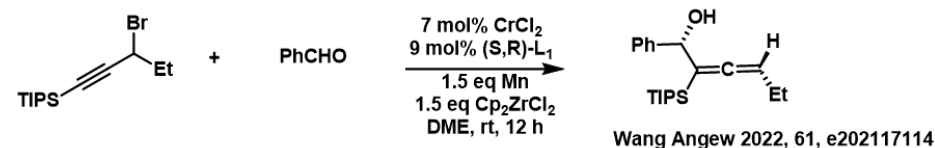
Research progress in recent years

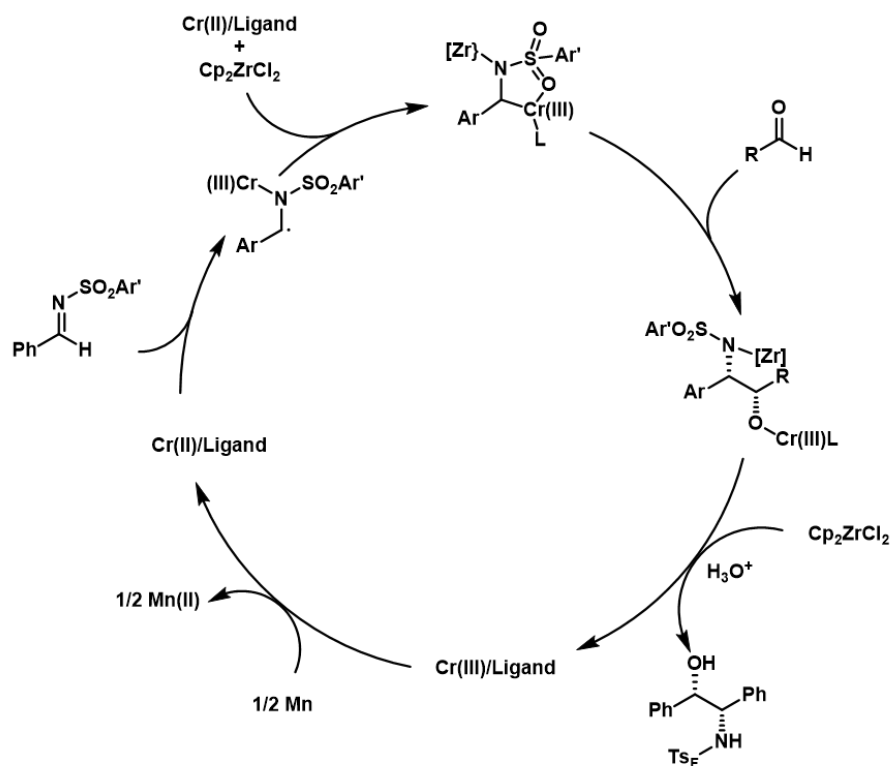
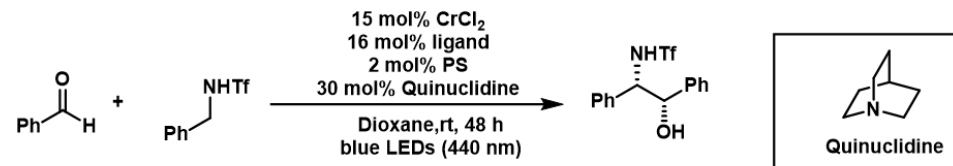


Baran JACS 2021, 143, 9478

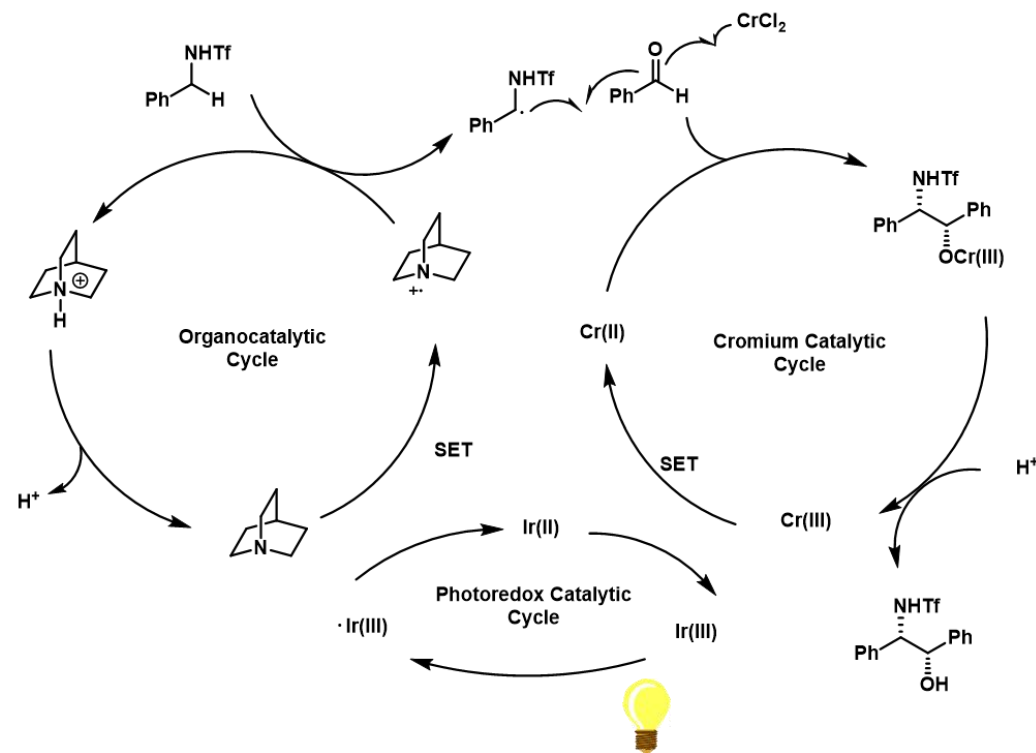


Glorius JACS 2020, 142, 2168

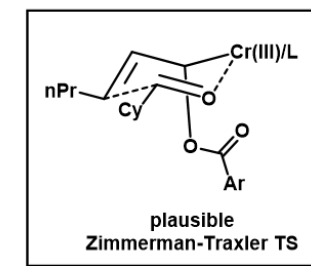
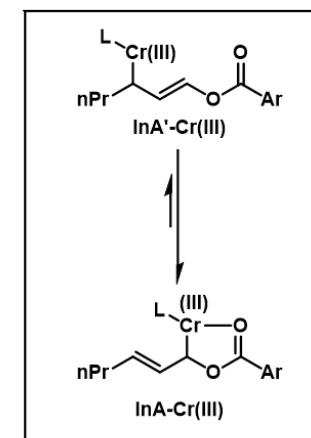
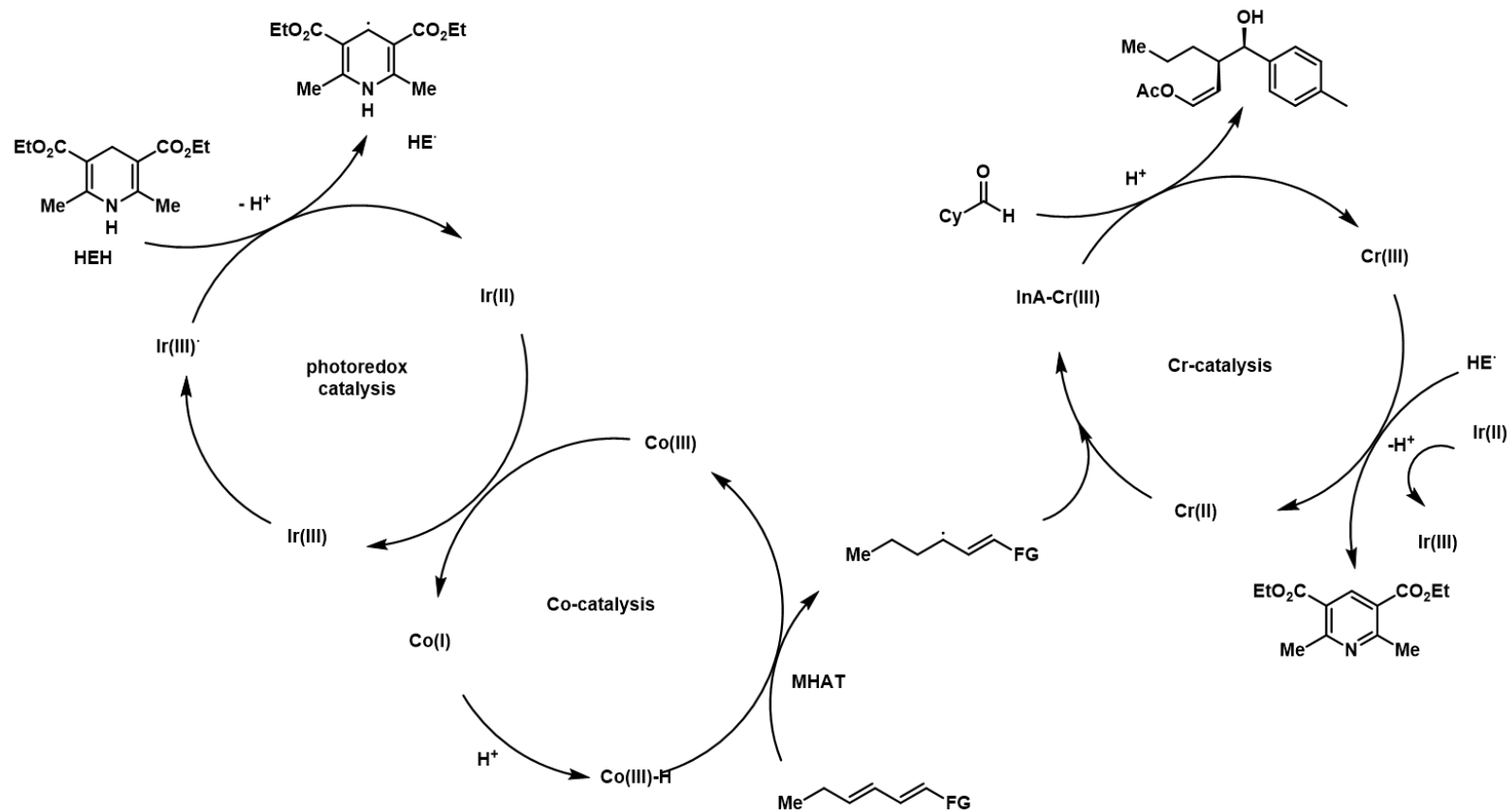
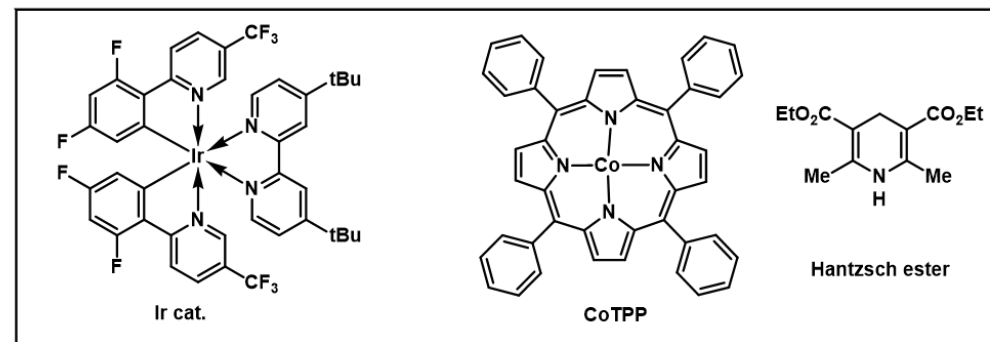
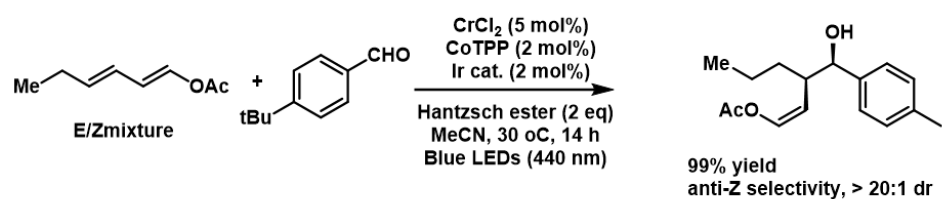




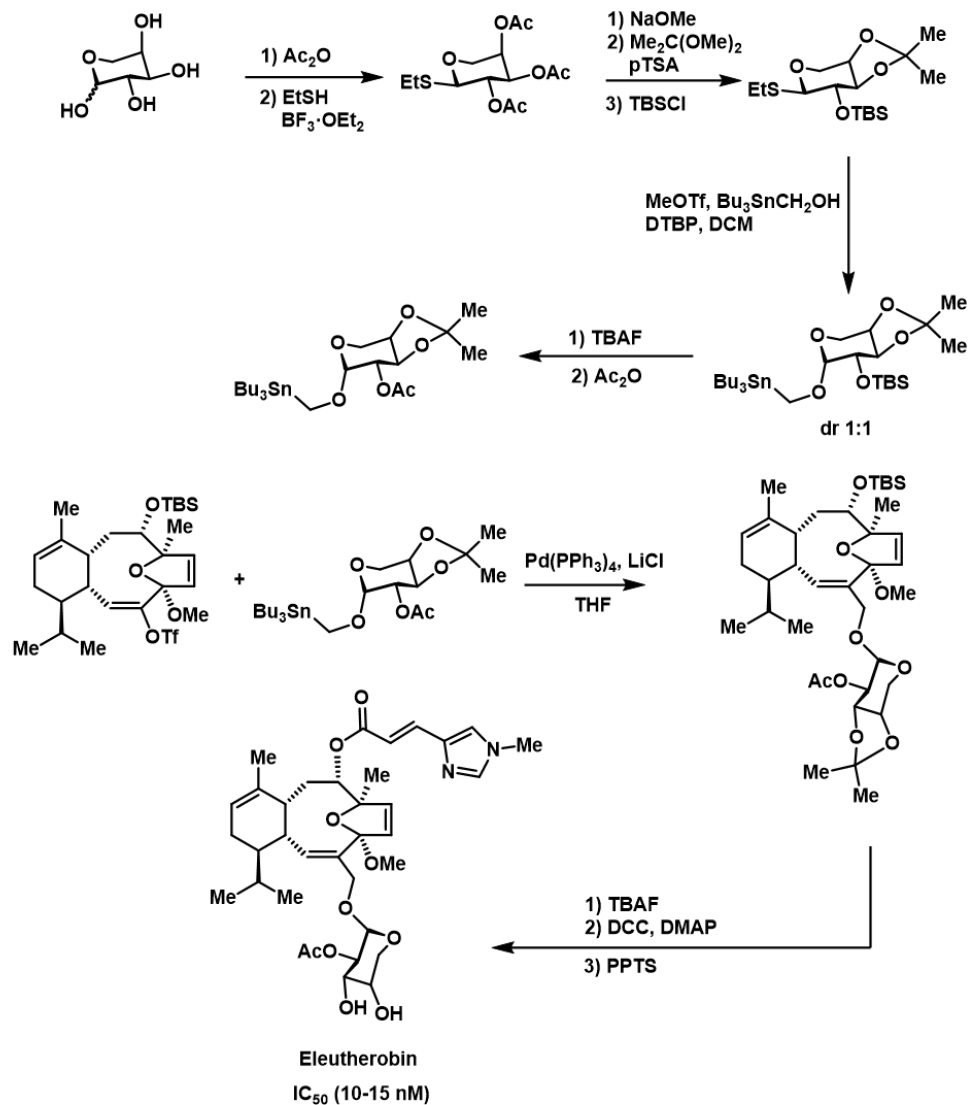
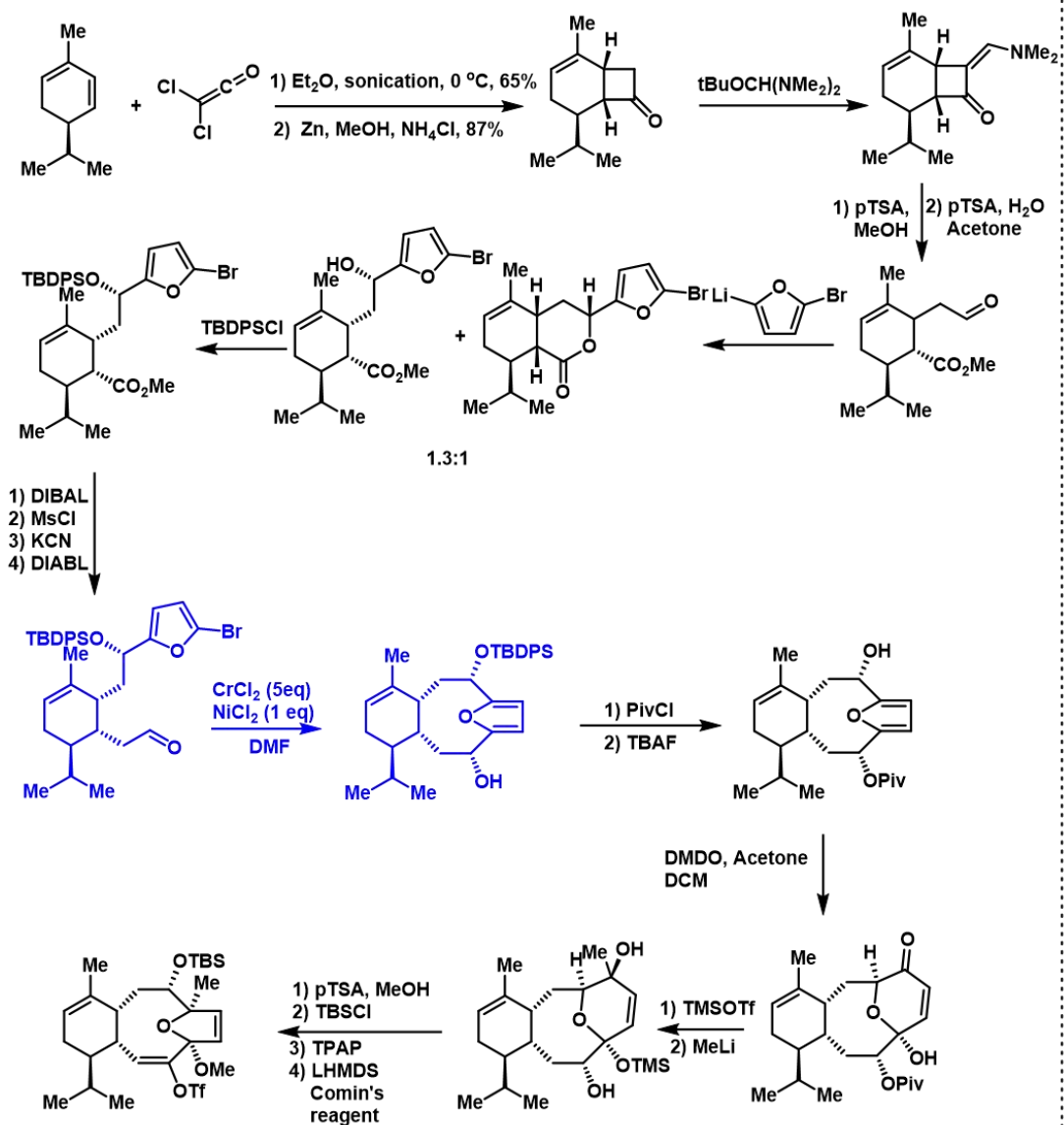
Wang JACS 2023, 145, 38, 20775

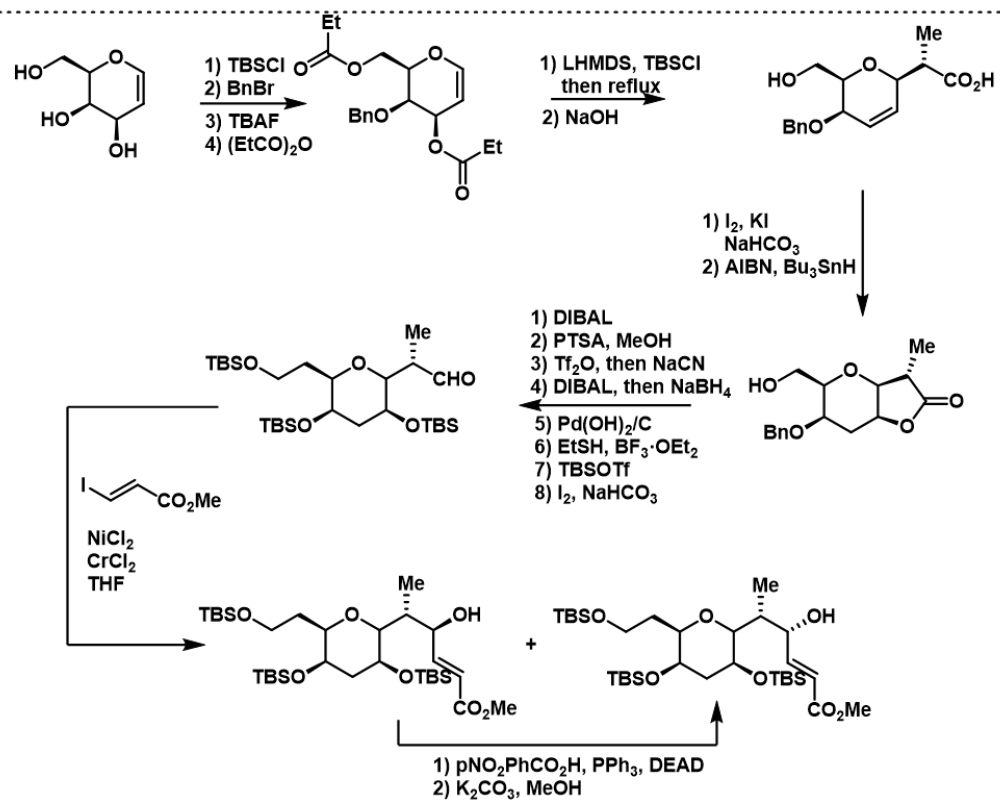
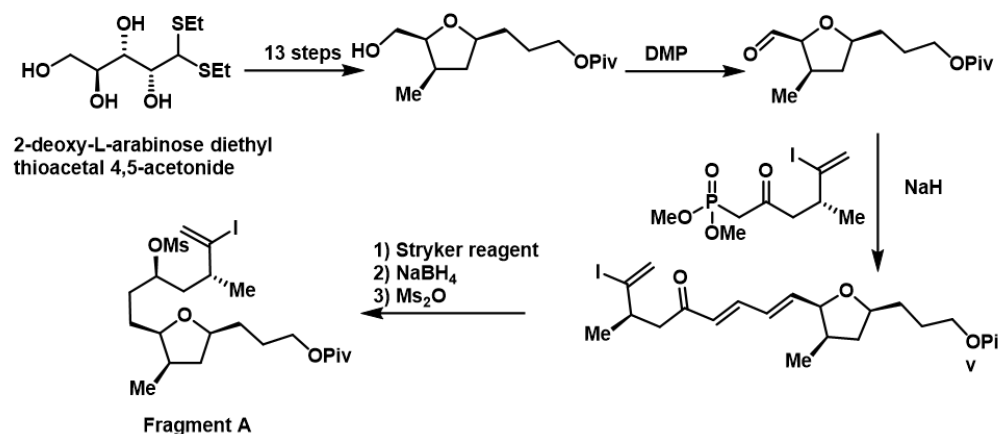
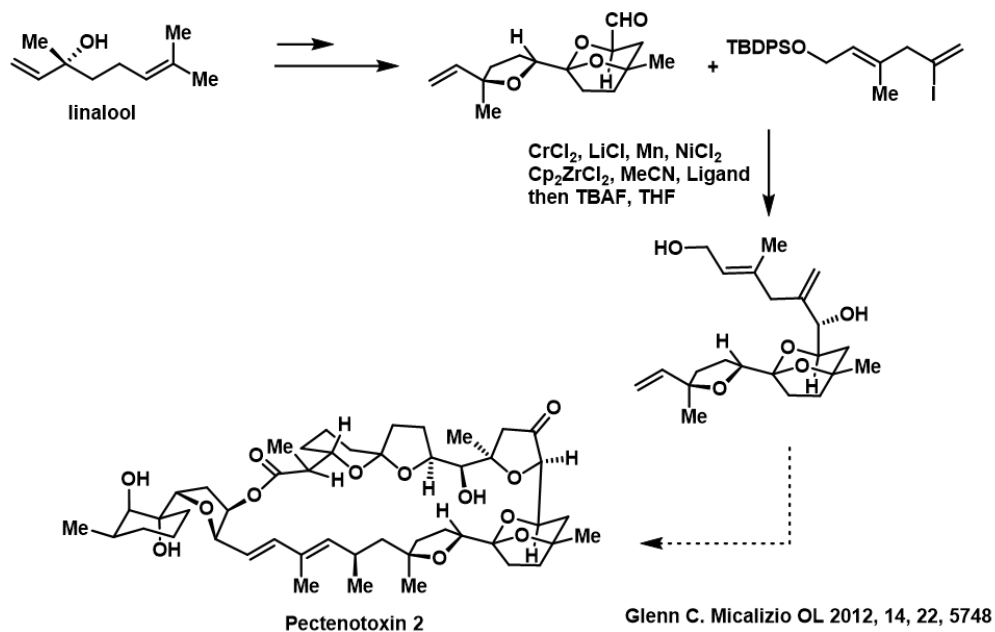
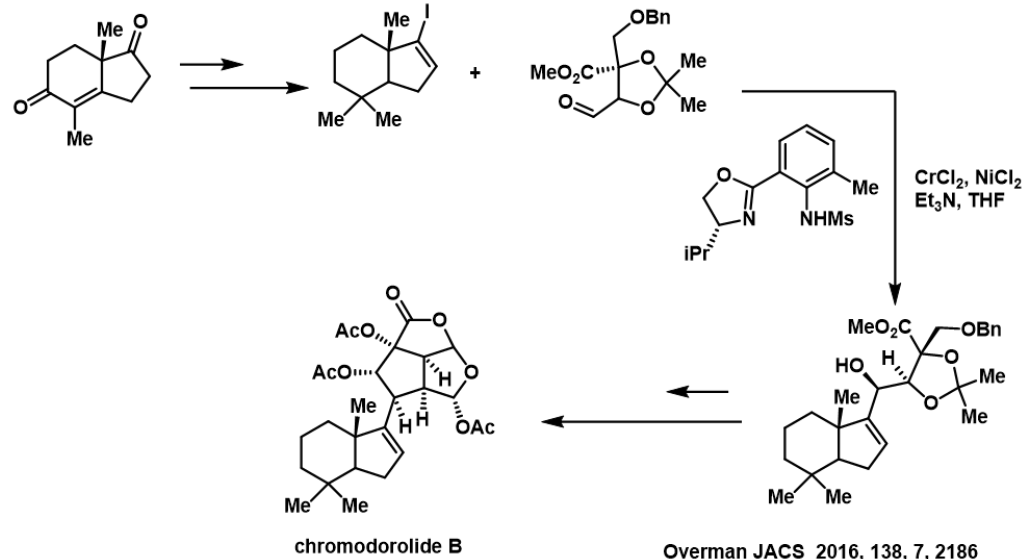


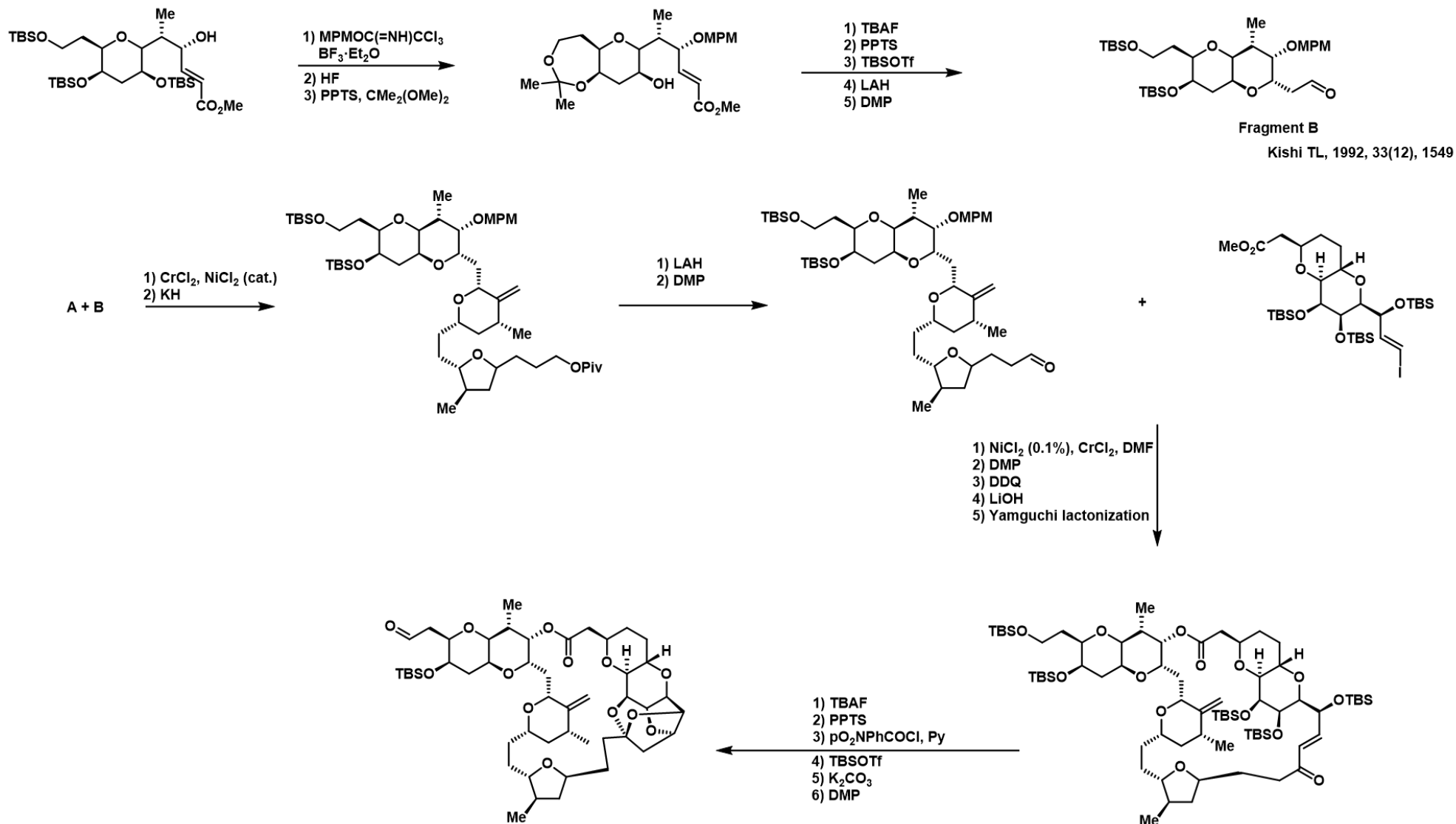
Wang JACS 2024, 146, 5316

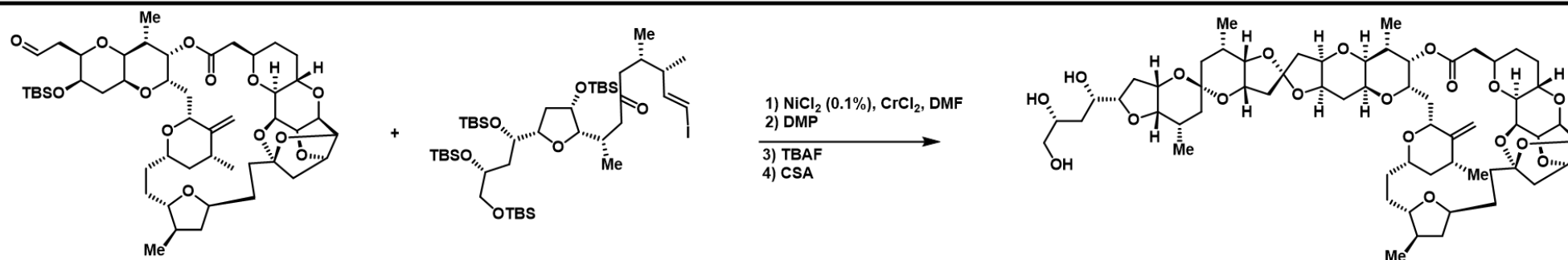


Synthetic application:



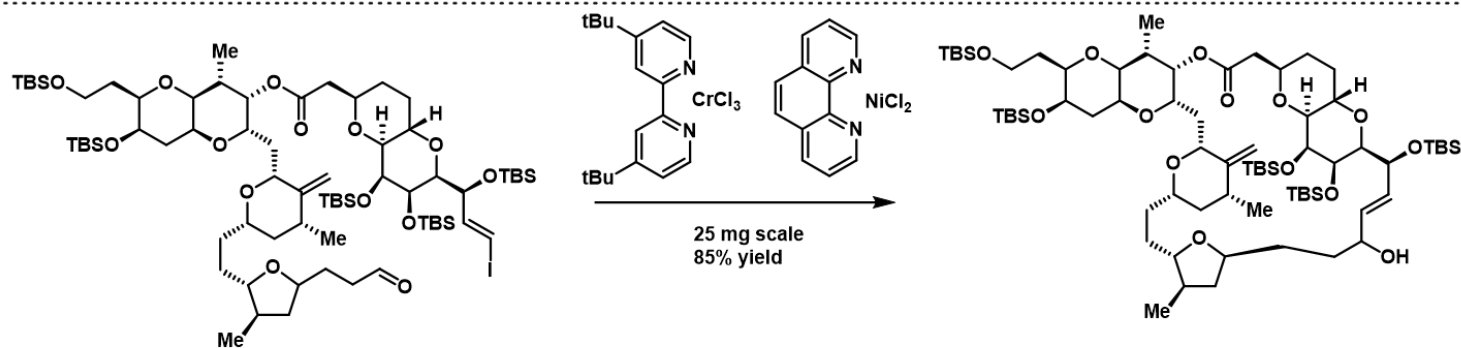






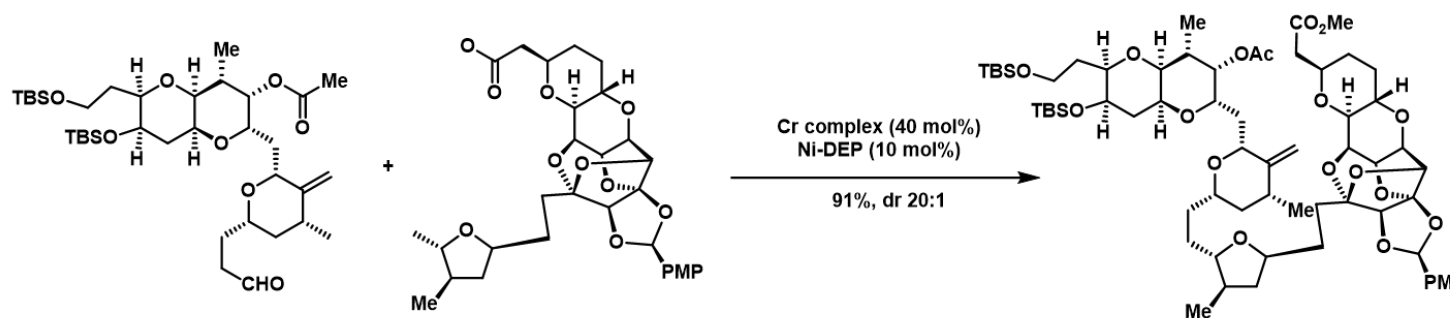
Halichondrin B

Kishi JACS 1992, 114, 3162

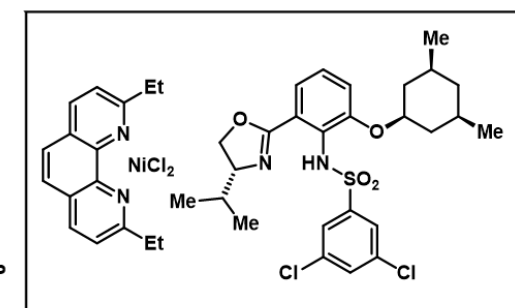


Catalytic Ni/Cr-Mediated Macrocyclization without Use of High-Dilution Techniques

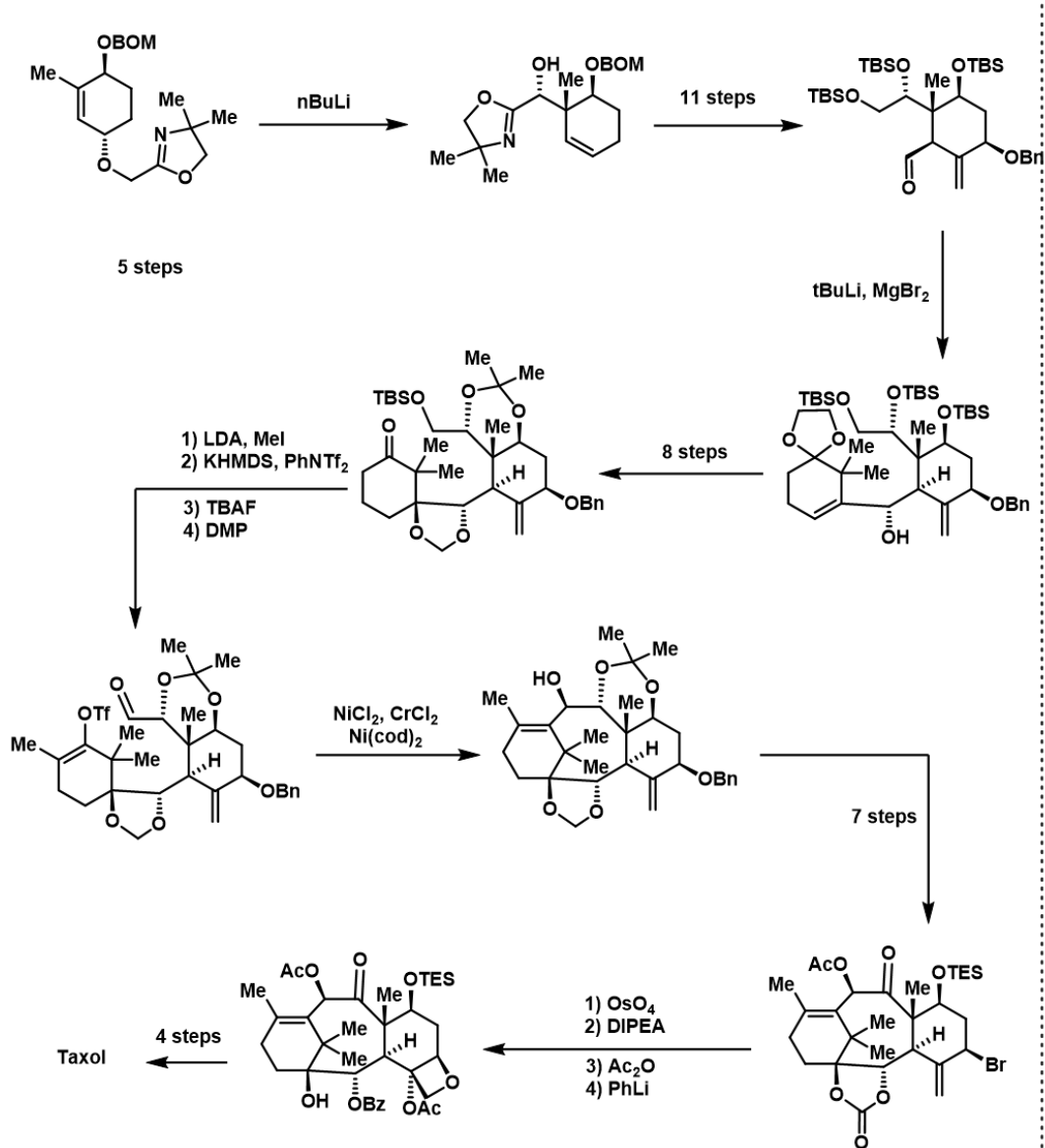
Kishi JACS 2005, 127, 15382



Total Synthesis of Halichondrin A

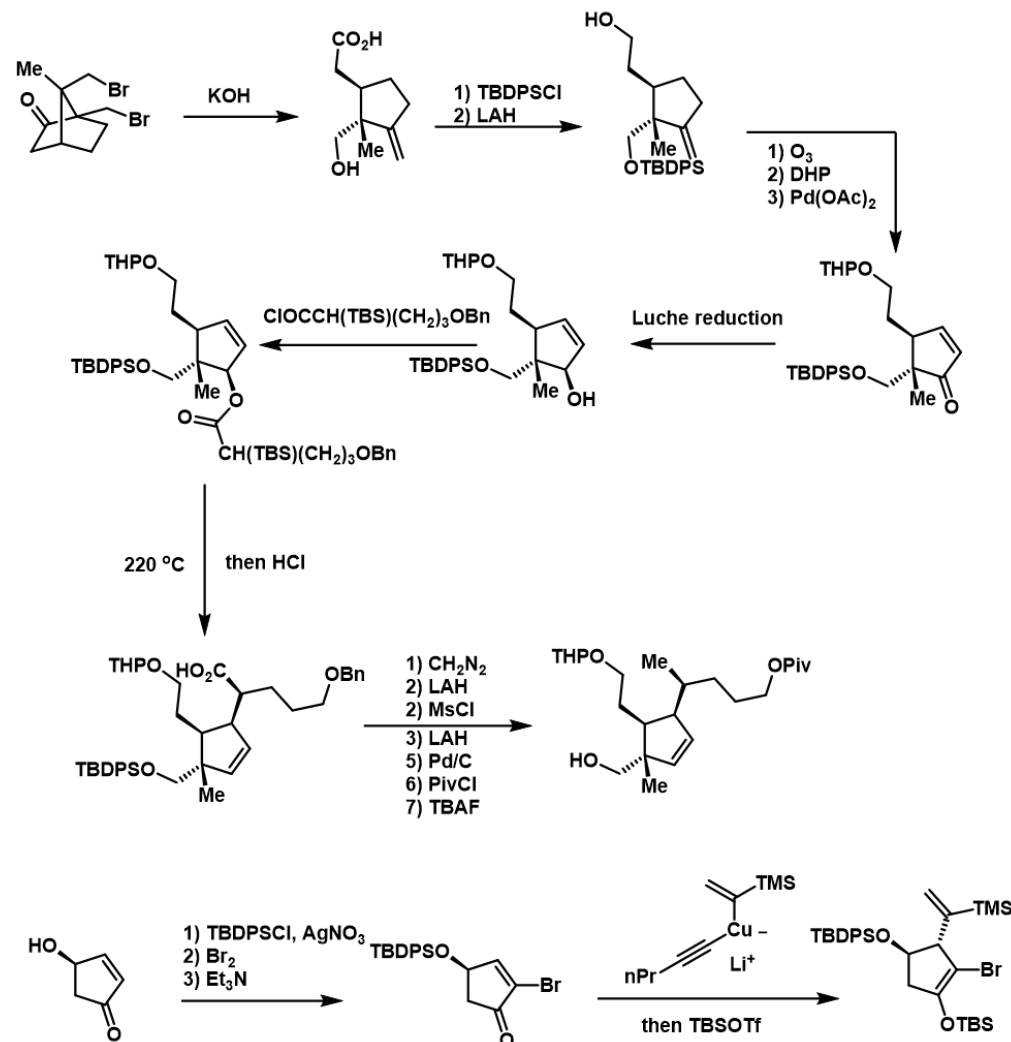


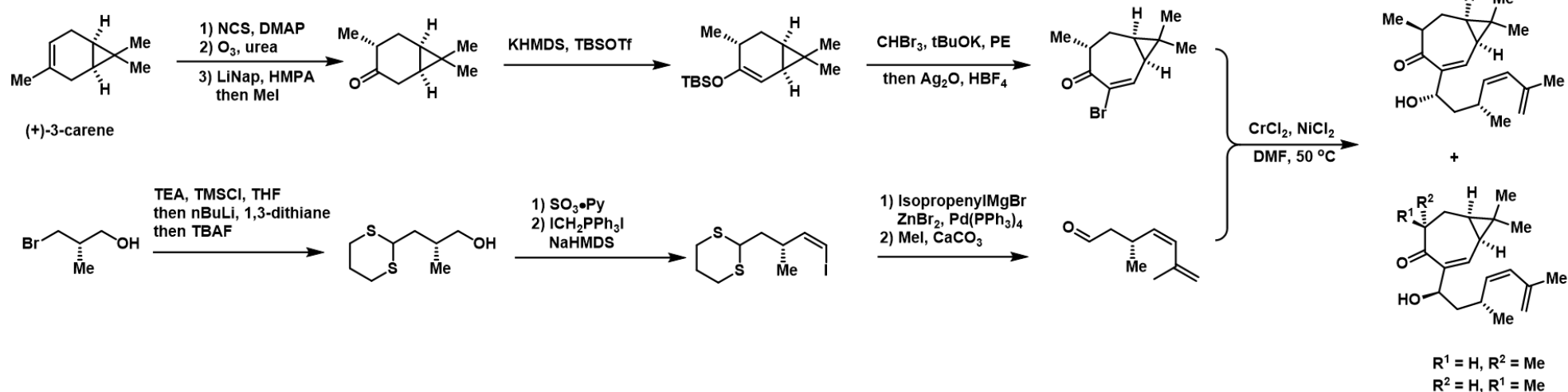
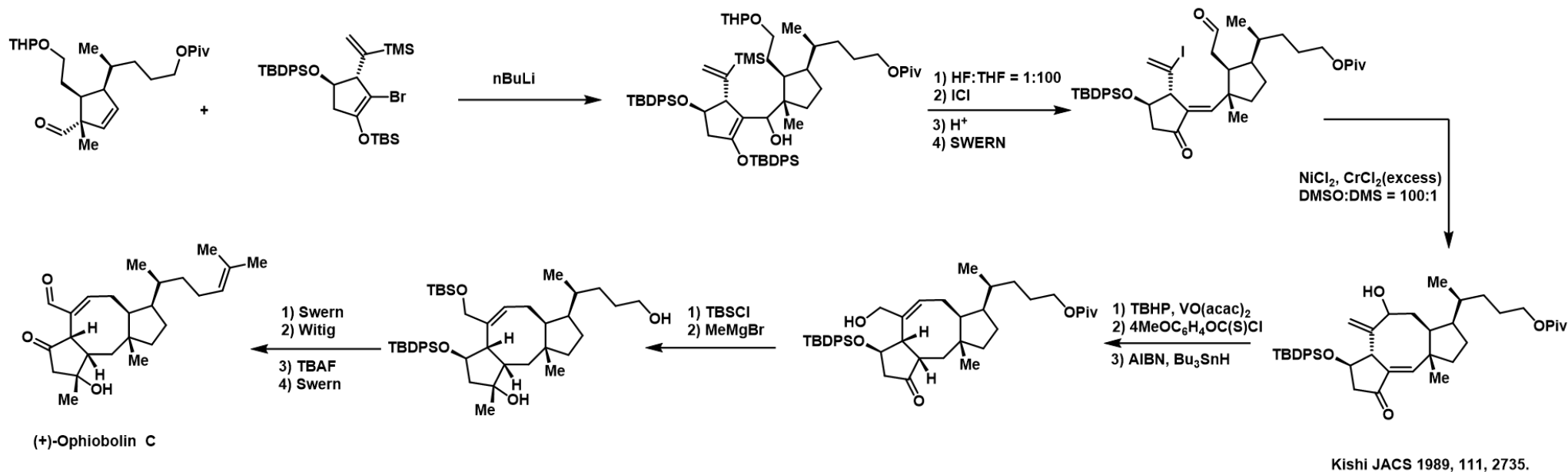
Kishi JACS 2014, 136, 5171.

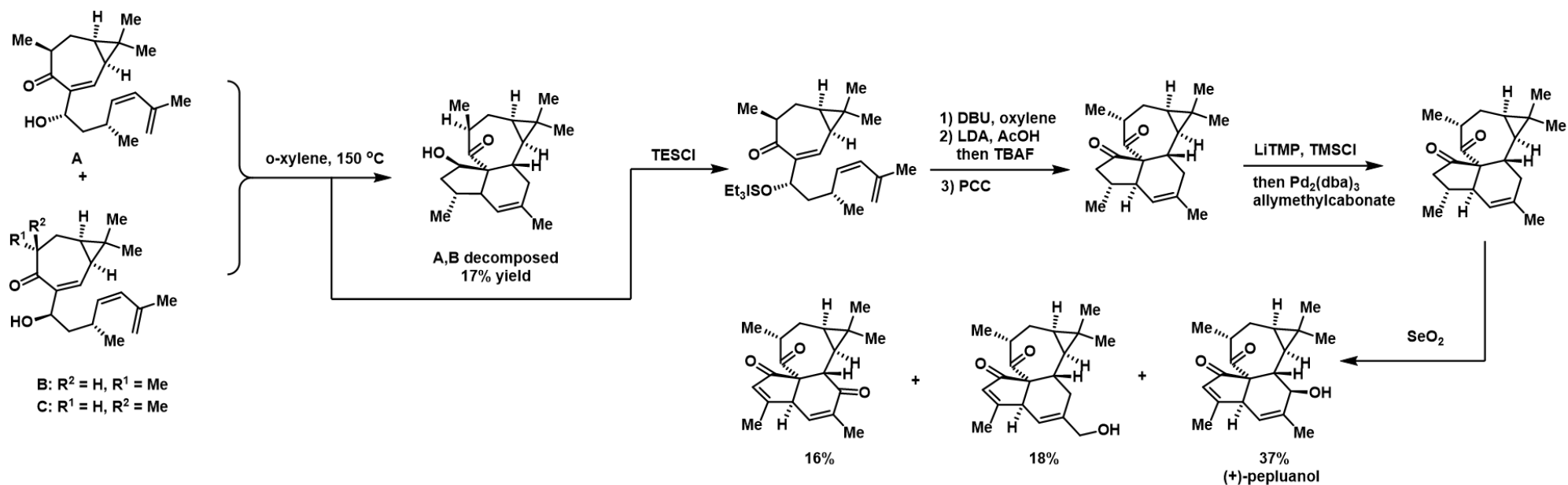


Kishi TL 1993, 34, 5999
Kishi TL 1993, 34, 6003
Kishi TL 1993, 34, 8047

Lim, J. Total Synthesis of Taxol. Ph.D. Thesis, Harvard University, 2000







Gaich JACS 2021, 143, 11934

